



HAL
open science

Enabling molecular communication through chirality of enantiomers

Valeria Loscri, Anna Maria Vegni

► **To cite this version:**

Valeria Loscri, Anna Maria Vegni. Enabling molecular communication through chirality of enantiomers. ITU Journal on Future and Evolving Technologies, 2021. ineris-03245796

HAL Id: ineris-03245796

<https://inria.hal.science/ineris-03245796>

Submitted on 2 Jun 2021

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

ENABLING MOLECULAR COMMUNICATION THROUGH CHIRALITY OF ENANTIOMERS

Valeria Loscri¹ and Anna Maria Vegni²

¹Inria Lille - Nord Europe, ²Roma Tre University, Rome, Italy

Abstract – With the advancements of nanotechnology, there has been fervid research activity on new communication paradigms suitable for new challenging contexts, such as biological systems. Among different approaches, the most considered has been the artificial Molecular Communication, where entities such as synthetic molecules, enzymes, hormones, bacteria etc. are functionalized in order to implement information exchange with the surrounding system and with other entities. In this context, it is interesting to analyze specific features that could be exploited for effective communication paradigms. In this paper, we focus on chiral molecules (a.k.a. enantiomers) as novel enablers for molecular communication paradigm. Chirality is an interesting and appealing feature existing in nature and that can be replicated with strong emphasis in new types of materials, such as metasurfaces and metamaterials. A deep knowledge of chirality features and how chiral molecules interact to each other or with achiral molecules gives insights into designing a new molecular communication technique suitable for biological environments. In this contribution, we will highlight the main applications of chiral molecules and we will present chiral features as the viable way for realizing a nanocommunication system.

Keywords – Chiral molecules, enantiomers, optical activity, chirality transfer, molecular communications

1. INTRODUCTION

Molecular Communication (MC) paradigm consists of using molecules to encode, transmit and receive information. It has recently received a lot of attention by the research community since it is considered as the viable alternative of electro-magnetic (EM) communications, thanks to the specific features of biocompatibility. MC is mostly inspired by existing communication mechanisms occurring between biological entities and is developed by considering small molecules, peptides, lipids, as well bacteria, viruses, pheromones and so on [2].

MC paradigm is based on the transmission and reception of information encoded into molecules (*i.e.*, messenger molecules) [1, 23]. These entities freely propagate in the medium by connecting a transmitter with a receiver nanomachine. Typical molecular communication systems are based on the free diffusion process of molecules, such as calcium signaling, microtubules, pheromone signaling, and bacterium-based communications [20]. Different biological entities allow reaching different communication ranges and performance. For instance, the use of pheromones (*i.e.*, molecules of chemical compounds released by plants, insects, and other animals) triggers specific behaviors among the receptor members and reaches long-range communications *i.e.*, approximately one meter. On the other side, both flagellated bacteria and catalytic nanomotors are able to carry DNA messages and allow short-range communications. The use of DNA as information messages allows achieving information rate relatively high (*i.e.*, up to several kilobits per second). In con-

trast, the propagation of information by means of guided bacteria or catalytic nanomotors is relatively very slow (*i.e.*, a few millimeters per hour).

A special type of molecules that is expected being very promising in the field of molecular communications are the *chiral molecules*. Chiral molecules show the chirality effect *i.e.*, a physical phenomenon that pervades the universe. The term chirality was introduced in 1884 and refers to objects that are not equivalent to their mirror images, and the two images are not superposed to each other. A typical example of such a geometrically chiral object is the human hand, so that the left and the right hands are mirror images of each other, but it is impossible to superpose them.

Chiral molecules, a.k.a. *enantiomers*, can show their different handedness in many ways, including the way they affect human beings. As an instance, one enantiomeric form of a compound called limonene is primarily responsible for the odor of oranges, while the other enantiomer, for the odor of lemons. Molecules of the amino-acids of which our proteins are built have the property of being non-superposable on their mirror image. In contrast, objects (and molecules) that are superposable on their images are achiral. The chirality of molecules can be demonstrated with relatively simple compounds. For instance, consider 2-butanol molecule *i.e.*, an organic compound with formula $\text{CH}_3\text{CH}(\text{OH})\text{CH}_2\text{CH}_3$. The structure of 2-butanol is chiral, that means that there are actually two different 2-butanols and they are enantiomers. Another example is the amino-acid alanine, which is in the

61 form of left-handed and right-handed enantiomers *i.e.*,
 62 (S)-alanine and (R)-alanine, respectively. Fig. 1 depicts
 63 the two enantiomers, whose mirror images are not equiv-
 64 alent and non-superposable on each other. Another ex-
 65 ample of enantiomers are the molecules of natural sugars,
 66 almost all classified as being right handed, including the
 67 sugar that occurs in DNA. Also DNA is a chiral structure,
 68 since its two helices are not superposable to each other,
 69 as well depicted in Fig. 1 (b).

70 In this paper we address how to exploit chiral molecules
 71 as messenger molecules for molecular communications.
 72 The use of similar molecules *i.e.*, isomers, as enablers
 73 of molecular communications has already been investi-
 74 gated in [15], where Kim and Chae proposed three
 75 novel modulation techniques, *i.e.*, concentration-based,
 76 molecular-type-based, and molecular-ratio-based. How-
 77 ever, in [15] it did not emerge the main features of such
 78 special molecules and how it is possible to exploit them
 79 for molecular communications by means of their inner
 80 features. As an instance, one of the main features of chiral
 81 molecules is their behavior towards plane-polarized light.
 82 When a beam of plane-polarized light passes through an
 83 enantiomer, the plane of polarization rotates. Also, sepa-
 84 rate enantiomers rotate the plane of plane-polarized light
 85 equal amounts but in opposite directions. Then, separate
 86 enantiomers are optically active compounds.

87 In this paper, we focus on the features of chiral molecules
 88 such as (i) the rotation of the polarization plane of the
 89 impinging EM wave and (ii) the chirality transfer effect,
 90 in order to model a chiral channel comprised of enan-
 91 tiomers that forward data information via a multi-hop
 92 protocol. Specifically, in our vision, data information is
 93 represented by exploiting the chirality phenomenon with
 94 a chiral molecule emitting a rotated EM wave when a
 95 light input impinges on an initial transmitter node after
 96 a steady-state is achieved. Dissemination of data infor-
 97 mation inside a chiral medium occurs through the *chiral-*
 98 *ity transfer* mechanism that considers the non-covalent
 99 bonds between a chiral and an achiral molecule. Chiral-
 100 ity is exploited to encode the information in the chiral
 101 molecules, and it is decoded at the receiver as a 1 bit when
 102 an EM wave (*i.e.*, an optical signal) is applied. When no EM
 103 wave is applied, the information is decoded as a bit 0.

104 This paper is organized as follows. Section 2 intro-
 105 duces the concept of chirality effect and describes the
 106 main enantiomers that can be found in nature, specifi-
 107 cally in the biological context. The features of chiral
 108 molecules are then presented in Subsection 2.1. In Sec-
 109 tion 3 we characterize the chiral transfer effect from a
 110 chiral molecule to an achiral molecule. In Section 4, we
 111 define a chiral medium as a channel for molecular com-
 112 munications. Data information is encoded in the chiral
 113 molecules that transport the chirality effect, which can be
 114 forwarded hop-by-hop in the overall system. Finally, con-
 115 clusions are drawn at the end of this paper.

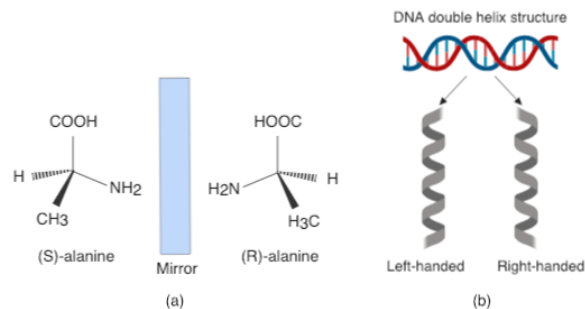


Fig. 1 – Examples of chiral molecules in case of (a) alanine amino-acid, and (b) DNA chains. Enantiomers are non-superposable mirror images to each other.

116 2. CHIRAL MOLECULES

117 A chiral molecule shows the chirality effect that makes the
 118 molecule not equivalent to its mirror image. The chiral
 119 molecule and its mirror image are enantiomers, and the
 120 relationship between the chiral molecule and its mirror
 121 image is defined as enantiomeric. The word chiral comes
 122 from the Greek, and means “hand”. Indeed, a classic ex-
 123 ample of chiral objects are the hands, since the mirror im-
 124 age of the left hand is exactly the right hand. However, the
 125 left hand is not superposable on the right hand.

126 From the etymology of chiral, chiral objects are said to
 127 possess “handedness”. Although both mirror image forms
 128 are theoretically possible, such as those for the amino-
 129 acid alanine, they have evolved in a way that amino-acids
 130 are mainly of the mirror image said to be “left-handed”
 131 (see Fig. 1 (a)). The reason that most amino-acids are of
 132 the left-handed form is not known, however.

133 Chirality is an important phenomenon in the universe.
 134 Several plants show chirality, by winding around support-
 135 ing structures. Human body is structurally chiral and it is
 136 not clear why, but most people are right-handed screw.
 137 Usually, only one form of chiral occurs in a given species.
 138 Just as an example, the molecules of white sugars are right
 139 handed. Enantiomers of a chiral molecule have identi-
 140 cal physico-chemical properties, and also the same elec-
 141 trochemical behavior. The enantioselective electrochem-
 142 istry represents the ability of discriminating enantiomers
 143 of chiral molecules (*i.e.*, electroanalysis), or to selectively
 144 activate or achieve a given enantiomer of a chiral molecule
 145 (*i.e.*, electrosynthesis) and is an issue particularly impor-
 146 tant in the biological and pharmaceutical fields [3].

147 2.1 Features of Chiral Molecules

148 The specific rotation is a property of a chiral molecule. It
 149 is defined as the change in orientation of monochromatic
 150 plane-polarized light, per unit distance-concentration
 151 product, as the light passes through a sample of a com-
 152 pound in solution. Fig. 2 describes the property of ro-
 153 tation of the polarization plane of an EM field that im-
 154 pinges on a chiral channel. At the output of the channel,
 155 the polarization plane has been rotated. Specifically, chi-
 156 ral molecules can rotate the plane of polarization of an EM

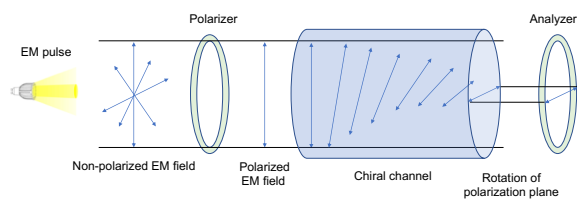


Fig. 2 – Property of rotation of the polarization plane of an EM pulse impinging a chiral channel.

field both clockwise and counterclockwise. If the rotation is clockwise, chiral molecules are said to be *dextrorotary*, and correspond to positive rotation values, while chiral molecules rotating the plane of polarization counterclockwise are said to be *levorotary*, and correspond with negative values of the specific rotation.

Another important feature of chiral molecules is the *chirality transfer* that occurs when a chiral molecule encounters an achiral molecule. In such a scenario, the chirality effect is extended over the whole molecular system *i.e.*, it propagates from a chiral to an achiral molecule, which becomes chiral. The induction of the chirality in the achiral components is of utmost importance. In order to induce the chirality of the achiral components, the interaction between the chiral molecules and the achiral molecules plays a very important role.

The induced chirality generally refers to those chiral supramolecular systems where chirality is induced in an achiral guest molecule as a result of asymmetric information transfer from a chiral host *e.g.*, a chiral molecule or a chiral nanostructure. In order to produce the induced chirality, it is necessary for the achiral molecule to have a strong interaction with the chiral host through a non-covalent bond. A typical example of induced chirality is the encapsulation of a chromophore into the cavity of cyclodextrin [6]. Finally, a very important aspect related to chirality is the *chiral communication* that is a common phenomenon occurring in many biological processes [10], strictly tied with the chirality transfer property.

In this paper, we will exploit the chirality transfer effect by the means of diffusion of chiral molecules entering in contact with achiral molecules in a biological solution.

3. CHIRALITY TRANSFER EFFECT

As already introduced in Subsection 2.1, the chirality transfer effect is observed between organic and inorganic molecular structures. The modeling of macroscopic chirality emerged from the chiral molecular elements is a challenge for theory, computations, and experiments. Numerous experimental results demonstrated the transfer of chirality among different length scales ranging from dimensions of the elementary particles to the macro-scale (*i.e.*, the length of the axon). In particular, it was shown that the chirality at the molecular scale (*i.e.*, amino-acids, proteins, and polysaccharides) could be transferred to the macroscopic and macrolevel (*i.e.*, neurofilaments and in-

organic crystals).

In general, chirality transfer occurs through chemical bonds, but recently it has been observed that chiral biomolecules may impart some of their optical properties to a spatially separated achiral dye [17]. Knof and von Zelewsky [16] have characterized the chiral transfer as through the use of organic ligands chiral information can be transferred.

In the context of chirality transfer it is important to highlight chiroptical properties. Among the most important properties are the chiral luminescent lanthanide complexes used as probes for the characterization of chiral environments [22, 4, 25] or as chiral luminescent complexes [21, 8, 9, 5]. Lanthanides are ideal candidates as luminescent probes, based on specific features such as their long lifetimes and large Stokes shifts. CP luminescence has great potential to investigate the configurational as well as conformational changes in biological systems in solution, since it combines the general sensitivity of luminescence measurements and the high specificity of the signal for the chiral environment. Furthermore, using very simple chiral ligands and lanthanide ions, chiral nanoballs were obtained where the array of lanthanide ions are arranged as in the ferritin biological molecule [13, 14]. With very simple chiral ligands used as synthons (*i.e.*, a synthon is a component of a molecule to be synthesized, playing an active role in synthesis) in coordination chemistry, it is possible to obtain sophisticated chiral assemblies which mimic biological systems. These considerations are encouraging in the development of “artificial” molecular communication systems.

Chirality is also important in molecular switches [12, 19]. A switch is a molecule that can reversibly interconvert between two stable states upon an external stimulus. In [11] Dai *et al.* described a chiroptical switch based on photochromes that exhibit two different states with significantly different optical rotations. Finally, chiral transfer phenomena can be used for sensing chirality of a wide range of chiral molecules, as well as for developing novel chiroptical devices and chiral materials.

The wider application of chiral sensing continues to be hampered by the involved chiral signals being inherently weak. To avoid this issue, plasmonic and dielectric nanostructures have recently been shown to offer a viable route for enhancing weak circular dichroism (CD) effects. Recently, in [18] Mohammadi *et al.* presented an analytical study of the problem of substrate CD spectroscopy for an arbitrary nanophotonic substrate (either, chiral or achiral, plasmonic or dielectric) and clarify the interplay between key affecting parameters, such as the thickness and chirality of the substrate, as well as the near-field optical chirality enhancement.

From the telecommunication point of view, the chirality transfer can be exploited by the means of a diffusion process of the chiral molecules.

When the “chirality effect” is transferred to an achiral molecule and an optical signal is applied, it will be able to show an optical activity, and becomes a chiral molecule.

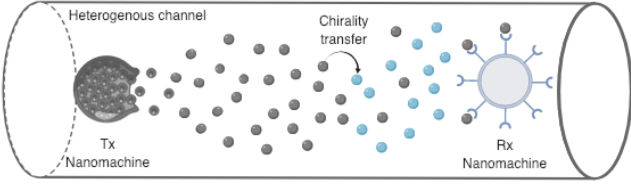


Fig. 3 – Chirality transfer property of chiral molecules (black circles) towards achiral molecules (light blue molecules).

As a result, the chirality transfer works as a point-to-point data forwarding among heterogeneous molecules (*i.e.*, from chiral to achiral molecules).

Specifically, when a chiral molecule encounters an achiral component, it forms a non-covalent bond and the chirality effect is transferred into the achiral molecule, which becomes chiral. Finally, the chirality transfer propagates in the whole system.

Fig. 3 depicts the chirality transfer feature of chiral molecules in a heterogeneous channel (*i.e.*, comprised of both chiral and achiral molecules). In this scenario, the chiral molecules are used as messenger molecules released by a transmitter nanomachine (*e.g.*, an eukaryotic cell) through the medium via diffusion (*i.e.*, Brownian motion). They are used as messengers, since they allow the transmission of a light signal applied to a transmitter molecule and transferred from a molecule to the neighbors through the chiroptical properties when the system will reach a steady state. The motion is basically driven by diffusion, meaning that the particles move from areas of higher concentration to areas of lower concentration, and the displacement of messenger molecules follows a normal distribution with zero mean.

The overall chiral molecule concentration flux is given by the sum of the N chiral molecules concentration gradients, with N as the number of apertures of the Tx nanomachine. The flux of chiral molecule concentration depends on both time and position through the Fick's first law *i.e.*,

$$J(x, t) = -D \sum_{i=1}^N \nabla C_{i,CM}(x, t), \quad (1)$$

where ∇ is an operator used in vector calculus as a vector differential operator, $C_{i,CM}$ [mol/cm³] is the i -th chiral molecule concentration with $i = \{1, 2, \dots, N\}$, and D [cm²/s] is the diffusion coefficient, assumed as a constant value for a given fluidic medium as:

$$D = \frac{k_B T}{3\pi\eta d}, \quad (2)$$

where k_B is the Boltzmann constant equal to 1.38×10^{-23} [J/K], T is the temperature [K], η is the viscosity of the liquid [mPa·s], and d is the size of the chiral molecules expressed in [nm]. Finally, Eq. (1) can be rewritten as:

$$J(x, t) = \frac{Q_{CM} + Q_{AM} Pr(AM \rightarrow CM)}{\sqrt{(4\pi Dt)^3}} e^{-\frac{x^2}{4Dt}} \quad (3)$$

where Q_{CM} is the initial concentration of chiral molecules, Q_{AM} is the initial concentration of achiral molecules and $J(x, t)$ represents the Brownian particles at time t at point x , with first moment as:

$$x^2 = 2Dt, \quad (4)$$

and standard deviation:

$$\sigma = \sqrt{2Dt}. \quad (5)$$

In Eq. (3), we account for the achiral molecules that are “inducted” to become chiral with a certain probability that is proportional to the helical twisting power of the chiral molecules, *i.e.*:

$$Pr(AM \rightarrow CM) \sim \beta, \quad (6)$$

where β is the helical twisting power and is expressed as [26]

$$\beta = \frac{\Delta\mu}{8\pi K_2 k}, \quad (7)$$

where $\Delta\mu$ is the chemical potential difference between a chiral molecule and its enantiomer when they are placed in the solution, K_2 is twist elastic constant, the wavevector $k = 2\pi/P$ and P is the elical pitch, which is inversely proportional to the concentration of chiral molecules injected by the transmitter.

4. CHIRAL OPTICAL CHANNEL

In the context of molecular communication, we envision that chiral molecules will be expected to be largely exploited [27]. Due to the feature of changing the polarization plane of an impinging optical signal, data information can be encoded into chiral molecules, and carried out via the chiral transfer mechanism. Specifically, when an optical pulse impinges a (biological) chiral channel, it will be observed an optical activity as output of the channel. The optical activity is expressed as a rotation of the polarization plane of the impinging EM wave. On the other side, if no pulse impinges the chiral channel, no optical activity will be observed at the output of the chiral channel, and then, there will be no rotation of the polarization plane of the EM wave.

In Fig. 4 we show how the chiral molecules are arranged after the diffusion process, in a steady-state. We assume that a certain concentration of chiral molecules is injected in the system and these molecules diffuse in the solution and “transfer” their chirality to other achiral molecules. Specifically, a Tx nanomachine releases a concentration of chiral molecules that transfer chirality to neighboring achiral molecules, which become chiral as well. Blue circles represent chiral molecules (both enantiomers), while grey circles are the achiral molecules. In this work, we treat chiral molecules as chiral optical antennas and we focus on some specific parameters allowing the characterization of chiral optical field generated. In particular,

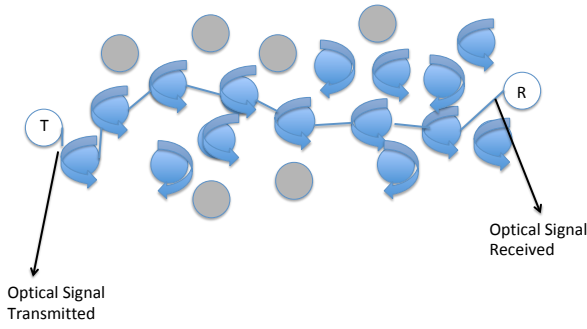


Fig. 4 - Representation of the chiral molecular channel after the diffusion process. The blue (gray) circles represent chiral (achiral) molecules.

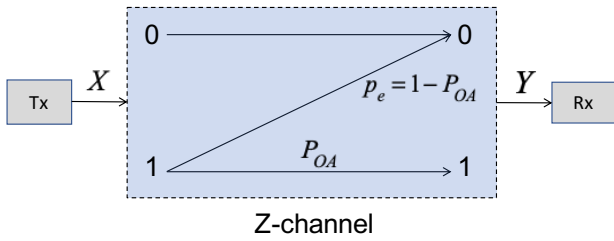


Fig. 5 - Z-channel model for the chiral medium.

as demonstrated in [24], we consider the chirality flux efficiency, that describes the ability of our molecules to scatter chiral optical fields, that will be considered as the input for the neighbor chiral molecule as shown in Fig. 4. In particular, the chiral flux efficiency can be written as [24]:

$$\eta_{\bar{F},d} = \text{Re}(\alpha_c) \frac{3}{8\mu_0} \frac{|\varepsilon_0|^2}{P_{tot}} \frac{1}{\sqrt{r^2 + \zeta^2}}, \quad (8)$$

with r and ζ as radial and longitudinal cylindrical coordinates, μ_0 and $|\varepsilon_0|^2$ the amplitude of the complex electric field in the point 0 and P_{tot} is the total power of the outgoing light. Most important is the parameter α_c representing the coupled magneto-electric polarizability, to whom chiral optical properties are attributed to. Of course, we have to consider that in molecules chiral optical signals are lower than in chiral metallic nanostructures, but there are recent research activities showing how it is possible to improve the quality of the signal [7].

The chiral channel can be then designed as a Z-channel, as depicted in Fig. 5. The output of the channel is expressed in terms of rotation of the polarization plane in case of an optical wave impinging the biological chiral channel.

From the telecommunications point of view, the optical activity due to the effect of chirality can be decoded as a 1 bit, while the absence of rotation of the polarization plane will be decoded as a 0 bit. Fig. 5 describes a chiral channel comprised of chiral molecules. A source node (*i.e.*, Tx node) emits a bit stream modulated through an On Off Keying (OOK) scheme. Specifically, the variable X rep-

resents the bit 1 or 0 transmitted along the channel, while Y is the received bit (*i.e.*, 1 or 0), based on a probabilistic approach. The presence of a bit 1 at the receiver means that a bit 1 has been transmitted with conditional probability $Pr(Y = 1|X = 1)$, while a bit 0 at the receiver side can be affected by errors in the channel corresponding to the error probability $p_e = Pr(Y = 0|X = 1)$, in case of a bit 1 transmitted with errors. Notice that, differently from traditional OOK-based communication schemes for molecular communications, in this paper the OOK modulation is not based on the concentration variation of the molecules, but on the optical activity generated by the chiral molecules. In practice, a bit 1 is associated to the optical activity occurrence, while a bit 0 is associated to no optical activity at the output of the channel.

Furthermore, we assume a time-based synchronization scheme at the transmitter side, and then the transmission of a bit 1 corresponds to an EM wave that excites the chiral channel at the beginning of a time slot, while no excitation corresponds to the emission of a bit 0. The probability of sending a bit 0 in the case of No Excitation at the beginning of the time slot is defined as P_{NE} . If no excitation is provided to the Tx node, then no optical activity will occur (*i.e.*, from the information theory point of view, no transmission errors will occur, while transmitting bit 0). The probability that the bit 1 is correctly received by the receiver corresponds to the probability of optical activity experienced by the chiral molecules, namely P_{OA} . This probability can be considered depending on the specific optical rotation *i.e.*, $[\alpha]_{\lambda}^T$, that is a physical constant of a chiral molecule, expressed as:

$$[\alpha]_{\lambda}^T = \frac{\alpha}{l \cdot \rho}, \quad (9)$$

where α is the optical rotation expressed in degrees, l is the optical path length [dm], and ρ is the concentration of sample in [g/mL], that we can derive from ((3)). In Eq. (9), we notice that the specific rotation depends on the wavelength λ [nm] of the impinging EM wave and the temperature T [Celsius]. Usually, the wavelength of the light used is 589 nm (*i.e.*, the sodium D line), and the symbol D is used *i.e.*, $[\alpha]_D^T$. The specific rotation can be either positive or negative, if the chiral molecules are dextrorotary or levorotary, respectively.

The probability of optical activity occurrence is intrinsically not equal to 1, *i.e.*, $P_{OA} \neq 1$, and can be expressed as:

$$P_{OA} = Pr \{ [\alpha]_{\lambda}^T > +0^\circ \}, \quad (10)$$

that means that if the specific rotation is greater than $+0^\circ$, then it is likely to have optical activity at the output of chiral channel. Notice that the specific rotation depends on the enantiomers that will rotate the plane of the polarized light of the same magnitude but in opposite directions (*i.e.*, $+$ or $-$). Without loss of generality, herein we assumed a positive rotation of the polarization plane, *i.e.*, we are assuming that the biological medium is comprised of a mixture of enantiomers ($+/-$) but the overall contribution of the specific rotation will be positive. On the

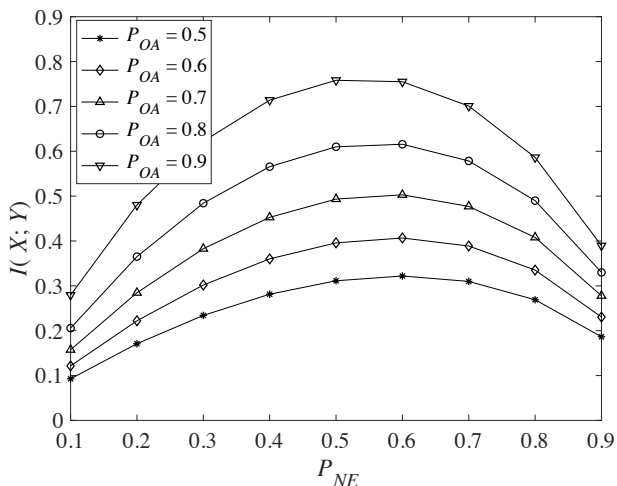


Fig. 6 – Mutual information related to the enantiomer Z-channel model.

other side, the channel will be optically inactive if the specific rotation will be null *i.e.*, there will be both 50%(+) and 50%(−) of enantiomers. This configuration is said racemate or racemic mixture.

According to the transmission probabilities of the Z-channel, the associated transition matrix is

$$\mathbf{P} = \begin{bmatrix} 1 & 0 \\ 1 - P_{OA} & P_{OA} \end{bmatrix}, \quad (11)$$

and then we can derive the mutual information between X and Y as:

$$I(X; Y) = H(P_{OA}(1 - P_{NE})) - (1 - P_{NE})H(1 - P_{OA}), \quad (12)$$

where $H(\cdot)$ represents the binary entropy. From Eq. (12), we can compute the channel capacity of the chiral medium as the maximum of the mutual information *i.e.*,

$$C_{chiral} = \max_{P_{OA}} [H(P_{OA}(1 - P_{NE})) - (1 - P_{NE})H(1 - P_{OA})], \quad (13)$$

where the probability distribution of input that maximizes the capacity will change according to specific chiral molecule concentrations.

Finally, Fig. 6 depicts the mutual information versus the probability of no excitation, for different values of P_{OA} . Notice that we consider only values of P_{OA} higher than 0.5, as it depends on the concentration of positive/negative enantiomers that comprise the channel. As expected, the highest value of the capacity (*i.e.*, 0.758 bit) is obtained for high value of P_{OA} .

5. CONCLUSIONS

This paper presents an overview of chirality effect in biomolecules. Chiral molecules *i.e.*, enantiomers, are present in nature everywhere, and therefore the main applications range from pharmaceutical to chemical and bi-

ological fields, including also communications and electronics. Apart natural enantiomers, artificial chiral materials (*i.e.*, chiral metamaterials) can be accordingly designed in order to exhibit an enhanced optical activity (*i.e.*, GOA effect).

In the context of communications, the use of natural chiral molecules, as well as chiral metamaterials, is envisioned as potential enablers for novel communication techniques. Specifically, in MC paradigm, chiral molecules have been analyzed as viable candidates for chiral communications, where information is encoded into chiral molecules. The features of rotation of the polarization plane and the chirality transfer have been exploited in order to derive a communication model based on chiral molecules. Information is represented by the optical activity, which can propagate inside a chiral medium by means of chirality transfer.

REFERENCES

- [1] I. F. Akyildiz, F. Brunetti, and C. Blázquez. Nanonetworks: A new communication paradigm. *Computer Networks*, 52(12):2260 – 2279, 2008.
- [2] I. F. Akyildiz, J. M. Jornet, and M. Pierobon. Nanonetworks: a new frontier in communications. *Communications of the ACM*, 54:84–89, November 2011.
- [3] S. Arnaboldi, M. Magni, and P. R. Mussini. Enantioselective selectors for chiral electrochemistry and electroanalysis: Stereogenic elements and enantioselection performance. *Current Opinion in Electrochemistry*, 8:60 – 72, 2018.
- [4] H. C. Aspinall. Chiral lanthanide complexes: Coordination chemistry and applications. *Chemical Reviews*, 102(6):1807–1850, 2002. PMID: 12059255.
- [5] J.-C. G. Benzli. Benefiting from the unique properties of lanthanide ions. *Accounts of Chemical Research*, 39(1):53–61, 2006. PMID: 16411740.
- [6] M. M. Bobek, D. Krois, and U. H. Brinker. Induced circular dichroism of cyclodextrin inclusion complexes: Examining the cavity with a bilateral probe. *Organic Letters*, 2(14):1999–2002, 2000. PMID: 10891214.
- [7] V. Bochenkov and T. Shabatina. Chiral plasmonic biosensors. *Biosensors (Basel)*, 8, 2018.
- [8] J. Bruce, D. Parker, S. Lopinski, and R. Peacock. Survey of factors determining the circularly polarised luminescence of macrocyclic lanthanide complexes in solution. *NCBI*, 14(7):562–7, Jul 2002.
- [9] J. Bruce, D. Parker, S. Lopinski, and R. Peacock. Synthesis and characterisation of highly emissive and kinetically stable lanthanide complexes suitable for usage “in cellulo”. *NCBI, Org. Biomol. Chem.*, 3(6):1013–24, Mar 2005.

- 453 [10] T. Chen, S. Li, D. Wang, M. Yao, and L. Wan. Remote
454 chiral communication in coadsorber induced
455 enantioselective 2d supramolecular assembly at a
456 liquid/solid interface. *Wiley Angewandte Chemie,
457 GDCh*, February 2015.
- 458 [11] Z. Dai, J. Lee, and W. Zhang. Chiroptical switches:
459 Applications in sensing and catalysis. *Molecules*,
460 17(2):1247–1277, 2012.
- 461 [12] B. L. Feringa, R. A. van Delden, N. Koumura, and
462 E. M. Geertsema. Chiroptical molecular switches.
463 *Chemical Reviews*, 100(5):1789–1816, 2000. PMID:
464 11777421.
- 465 [13] S. Ghosh and P. S. Mukherjee. Self-assembly of
466 molecular nanoball: Design, synthesis, and char-
467 acterization. *The Journal of Organic Chemistry*,
468 71(22):8412–8416, 2006. PMID: 17064013.
- 469 [14] K. S. Jeong, Y. S. Kim, Y. J. Kim, E. Lee, J. H. Yoon,
470 W. H. Park, Y. W. Park, S.-J. Jeon, Z. H. Kim, J. Kim,
471 and N. Jeong. Lanthanitin: A chiral nanoball en-
472 capsulating 18 lanthanum ions by ferritin-like as-
473 sembly. *Angewandte Chemie International Edition*,
474 45(48):8134–8138, 2006.
- 475 [15] N. Kim and C. Chae. Novel modulation techniques us-
476 ing isomers as messenger molecules for nano com-
477 munication networks via diffusion. *IEEE Journal on
478 Selected Areas in Communications*, 31(12):847–856,
479 December 2013.
- 480 [16] U. Knof and A. von Zelewsky. Predetermined chiral-
481 ity at metal centers. *Angewandte Chemie Interna-
482 tional Edition*, 38(3):302–322, 1999.
- 483 [17] X. Li and M. Shapiro. Communications: Spatial sep-
484 aration of enantiomers by coherent optical means.
485 *The Journal of Chemical Physics*, 132(4):041101,
486 2010.
- 487 [18] E. Mohammadi, K. Tsakmakidis, A. Askarpour,
488 P. Dehkhoda, A. Tavakoli, and H. Altug. Nanopho-
489 tonic platforms for enhanced chiral sensing. *ACS
490 Photonics*, 05 2018.
- 491 [19] E. Murguly, T. B. Norsten, and N. R. Branda. Nonde-
492 structive data processing based on chiroptical 1,2-
493 dithienylethene photochromes. *Angewandte Chemie
494 International Edition*, 40(9):1752–1755, 2001.
- 495 [20] T. Nakano, M. J. Moore, F. Wei, A. V. Vasilakos, and
496 J. Shuai. Molecular communication and networking:
497 Opportunities and challenges. *IEEE Transactions on
498 NanoBioscience*, 11(2):135–148, June 2012.
- 499 [21] D. Parker. Excitement in f block: structure, dynam-
500 ics and function of nine-coordinate chiral lanthanide
501 complexes in aqueous media. *NCBI, Chem. Soc. Rev.*,
502 33(3):156–165, Mar 2004.
- 503 [22] D. Parker, R. S. Dickins, H. Puschmann, C. Cross-
504 land, and J. A. K. Howard. Being excited by lan-
505 thanide coordination complexes: aqua species, chi-
506 rality, excited-state chemistry, and exchange dynam-
507 ics. *Chem Rev*, 102(6):1977–2010, February 2002.
- 508 [23] M. Pierobon and I. F. Akyildiz. A physical end-to-end
509 model for molecular communication in nanonet-
510 works. *IEEE Journal on Selected Areas in Communi-
511 cations*, 28(4):602–611, May 2010.
- 512 [24] L. V. Poulikakos, P. Thureja, A. Stollmann, E. D. Leo,
513 and D. J. Norris. Chiral light design and detection in-
514 spired by optical antenna theory. *ACS Nano Letters*,
515 18:4633–4640, 2018.
- 516 [25] H. Tsukube and S. Shinoda. Lanthanide complexes
517 in molecular recognition and chirality sensing of bi-
518 ological substrates. *Chem. Rev.*, 102(6):2389–2404,
519 Jun 2002.
- 520 [26] M. R. Wilson and D. J. Earl. Calculating the helical
521 twisting power of chiral dopants. *Journal of Materi-
522 als Chemistry*, 11:2672–2677, 2001.
- 523 [27] Y. Zhao, A. N. Askarpour, L. Sun, J. Shi, X. Li, and
524 A. Alú. Chirality detection of enantiomers using
525 twisted optical metamaterials. *Nature Communica-
526 tions*, 8, 2017.



528 **Valeria Loscrì** is a permanent researcher of the FUN
 529 Team at Inria Lille-Nord Europe since Oct. 2013. From
 530 Dec. 2006 to Sept. 2013, she was Research Fellow in
 531 the TITAN Lab of the University of Calabria, Italy. She re-
 532 ceived her MSc and PhD degrees in Computer Science in
 533 2003 and 2007, respectively, from the University of Cal-
 534 abria and her HDR (Habilitation a diriger des recherches)
 535 in 2018 from Université de Lille (France). Her research
 536 interests focus on emerging technologies for new commu-
 537 nication paradigms such as Visible Light Communication
 538 and TeraHertz bandwidth and cooperation and coexis-
 539 tence of wireless heterogeneous devices. She has been in-
 540 volved in the activity of several European Projects (H2020
 541 CyberSANE, FP7 EU project VITAL, the FP6 EU project
 542 MASCOT, etc.), Italian and French projects. She is in the
 543 editorial board of IEEE COMST, Elsevier ComNet, JNCA,
 544 IEEE Trans. on Nanobioscience. Since 2019, she is Sci-
 545 entific International Delegate for Inria Lille-Nord Europe.

561 cations. She a member of ACM and an IEEE Senior Mem-
 562 ber. In March 2018, she got the Italian Habilitation (Abili-
 563 tazione Scientifica Nazionale) for Associate Professorship
 564 in Telecommunication Engineering. She is involved in the
 565 organization of several IEEE and ACM international con-
 566 ferences and is a member of the editorial board of IEEE
 567 Communications Magazine, Ad Hoc Networks, Journal of
 568 Networks and Computer Applications, Nanocomnet Else-
 569 vier journals, IEEE JCN, ITU J-FET and ETT Wiley journal.



546 **Anna Maria Vegni** (Senior member, IEEE) is a tenure-
 547 track Assistant Professor in the Department of Engineer-
 548 ing at Roma Tre University (Italy), since March 2020.
 549 She received the Ph.D. degree in Biomedical Engineering,
 550 Electromagnetics and Telecommunications from the De-
 551 partment of Applied Electronics, Roma Tre University, in
 552 March 2010. She received the 1st and 2nd level Laurea
 553 Degree cum laude in Electronics Engineering at Roma Tre
 554 University, in July 2004, and 2006, respectively. In 2009,
 555 she was a visiting researcher in the Multimedia Commu-
 556 nication Laboratory, directed by Prof. Thomas D.C. Little,
 557 at the Department of Electrical and Computer Engineer-
 558 ing, Boston University, Boston, MA. Her research activity
 559 focused on vehicular networking supported by heteroge-
 560 neous wireless networks and optical wireless communi-