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LOW-MACH TYPE APPROXIMATION OF THE NAVIER-STOKES SYSTEM WITH TEMPERATURE AND SALINITY FOR FREE SURFACE FLOWS*

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6 Abstract. We are interested in free surface flows where density variations coming e.g. from temperature or salinity differences play a significant role in the hydrodynamic regime. In water, 8 acoustic waves travel much faster than gravity and internal waves, hence the study of models arising 9 from compressible fluid mechanics often requires a decoupling between these waves. Starting from 10 the compressible Navier-Stokes system, we derive the so-called Navier-Stokes-Fourier system in an "incompressible" regime using the low-Mach scaling, hence filtering the acoustic waves, neglecting 11 the density dependency on the fluid pressure but keeping its variations in terms of temperature and 12 13 salinity. A slightly modified low-Mach asymptotics is proposed to obtain a model with thermo-14mechanical compatibility. The case when the density depends only on the temperature is studied first. Then the variations of the fluid density with respect to temperature and salinity are considered, 15and it seems to be the first time that salinity dependency is considered in this low Mach limit. 16

Key words. Navier-Stokes equations, compressible and incompressible fluids, free surface flows,
 variable density flows, low-Mach approximation

19 AMS subject classifications. 35Q30, 35Q35, 76D05

20 NEW VERSION

4

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31 32

1. Introduction. In oceans and lakes, one of the predominant driving forces is the difference in density, caused by salinity and temperature variations: increasing the salinity and lowering the temperature of a fluid both increase its density [32]. Oceans and lakes are stratified: the water density varies along the vertical direction. In the present work, we aim at describing and simulating variable density flows with free surface.

The Navier-Stokes equations – or possibly simplified versions of these equations – are the cornerstone of the modeling of hydrodynamics in lakes and oceans. Compared to situations where the density is constant, when considering variable density flows with free surface, two characteristics appear:

- the variations of the fluid density can lead to acoustic waves,
- the fluid density adds nonlinearities in the Navier-Stokes equations.

33 These two aspects are detailed hereafter.

The density variations are usually small and a common assumption in the case of geophysical flows is the Boussinesq approximation [9]. It is widely used to simplify

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36 the Navier-Stokes equations with variable density and consists in ignoring density

37 variations in momentum conservation equations except in the buoyancy force term.
38 The underlying assumptions are 1) the density variations are very small with respect

39 to the mean density and 2) the vertical scale of the motion is small compared to the

40 vertical scale of variations in density, and the density and pressure variations due to

41 motion are not larger than the total static variations of pressure and density [36, 23]

42 The consequences of the Boussinesq approximation are listed in [4]. Notably, under

43 the Boussinesq approximation, the volume of the water is conserved, while the mass

44 is not conserved. With this approximation, the density can be defined as a function
45 of any given tracer, the temperature or a pollutant for instance. The Boussinesq
46 approximation is the basis of many ocean models such as ICON [21], NEMO [29] and

47 POM [31].

Several authors have shown the benefits of taking into account non-Boussinesq effects in lake and ocean models, either for the propagation of internal waves [35] or for sea level variations induced by expansion/contraction processes [28, 23, 16]. In the case of ocean water, density is in general a function of pressure, temperature and salinity. Equations of state for water based on experiments are proposed in [33, 38].

Consequently, approaches to take into account the non-Boussinesq effects have 53 been developed. A possibility is to adopt pressure coordinates [20, 34]. The non-54Boussinesq equations written in pressure coordinates are isomorphic to the Boussinesq equations in z-coordinates, which allows to use the same algorithm for the non-56 Boussinesq model as for the Boussinesq model. So far, however, this approach is not 58 widely used in ocean modeling, though it is available in the code MITgcm (|1|). A review of the different types of vertical coordinates used for ocean modeling is given in [18]. In [4], a non-hydrostatic non-Boussinesq model is presented. A non-hydrostatic 60 pressure anomaly is related to a compressible (non-Boussinesq) density anomaly. Yet 61 the authors of [4] are primarily concerned with the simulation of acoustic waves and 62 thermal dilation is not investigated numerically. This model has recently been in-63 64 cluded in CROCO ([11]). In the present work, we focus on the inclusion of dilation effects in the model. 65

The approach chosen here to propose a non-Boussinesq model is different from the 66 previous ones. Starting from the compressible Navier-Stokes equations, we propose a 67 formulation of the Navier-Stokes-Fourier system in the asymptotics of an "incompress-68 ible" fluid. The term "Fourier" refers here to Fourier's law of thermal conduction. For 69 70 most geophysical flows, the water can be considered as incompressible in the sense that the variation of its density with respect to the fluid pressure is small [33, 38] 71and in this case, the acoustic waves can be filtered out, this is the so called low Mach 72 number limit. This is advantageous from the computational point of view because a 73 74 restrictive condition must be imposed on the time step when the acoustic waves are included [30]. Even if in some very particular situations [22], a coupling between grav-75 ity and acoustic waves can occur in the context of fluid density variations, we consider 76 here situations (in terms of water depth and wave frequency) when the phenomena 77 are decoupled. 78

The incompressible limit of the Navier-Stokes-Fourier system has been extensively studied mathematically, see [2, 13] and the references therein. The incompressible limit is a low-Mach approximation of the Navier-Stokes-Fourier system. It consists in neglecting the pressure dependency in the fluid state law (but not the temperature and salinity dependencies). This limit is singular and requires not necessarily intuitive scalings, in particular for numerical approximations [8]. Since the studied phenomena strongly couple mechanics (fluid motion and rheology) with thermodynamics (temper-

ature and salinity variations), we pay close attention that the obtained models do not 86 87 violate the second principle of thermodynamics. The most simple way of performing the incompressible limit yields a model for which the energy balance exhibits discrep-88 ancies with respect to the energy balance of the original compressible system. To 89 recover an energy balance that is close to that of the original system, which is impor-90 tant for the robustness of numerical methods, some corrections terms are incorporated 91 in the incompressible model. These do not alter the accuracy of the approximation. Instead of conserving the volume of fluid and not the mass (which is what the Boussi-93 nesq approximation implies), the model derived here strictly conserves the mass and is 94enriched by the thermohaline expansion effects i.e. the volume is no longer conserved. 95 In particular the velocity is not divergence free, but the divergence of the velocity field 96 is equal to a right-hand side involving temperature and salinity gradients, as well as 97 viscous dissipation. Our incompressible models include the dependency of density on 98 salinity additionally to the dependency on temperature, and it seems to be the first 99 time that this is considered in detail, including Onsager's principles. The proposed 100 model is close to the one in [6], yet it is rigorously derived here. While solving the 101 model in [6] required inverting a non-linear system, the numerical resolution of the 102 103 present model is more simple, see [7]. Moreover, the model in the present work is 3D while it was only 2D in [6]. 104

105 A multilayer approximation of the Navier-Stokes-Fourier system with a numerical 106 scheme – endowed with strong stability properties – and several numerical test cases 107 will be presented in a forthcoming paper [7].

The paper is organized as follows. The incompressible Navier-Stokes-Fourier and Euler-Fourier models are introduced in Section 2 and their low Mach limits are studied in Section 3 including the situation where the temperature and the salinity are considered. Finally in Section 4, the formulation of the considered models with the hydrostatic assumption is given.

113 2. The 3d Navier-Stokes-Fourier system.

114 **2.1. The compressible Navier-Stokes-Fourier system.** We consider the 115 classical compressible Navier-Stokes system describing a free surface gravitational 116 flow over a bottom topography $z_b(x, y)$,

117 (2.1)
$$\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mathbf{U}) = 0,$$

118 (2.2) $\frac{\partial (\rho \mathbf{U})}{\partial t} + \nabla \cdot (\rho \mathbf{U} \otimes \mathbf{U}) + \nabla p - \nabla \cdot \sigma = \rho \mathbf{g},$
119 (2.3) $\frac{\partial}{\partial t} \left(\rho \frac{|\mathbf{U}|^2}{2} + \rho e \right) + \nabla \cdot \left(\left(\rho \frac{|\mathbf{U}|^2}{2} + \rho e + p - \sigma \right) \mathbf{U} \right) = -\nabla \cdot Q_T + \rho \mathbf{g} \cdot \mathbf{U},$

where $\mathbf{U}(t, x, y, z) = (u, v, w)^T$ is the velocity, ρ is the mass density, p is the fluid 120 pressure, σ is the viscosity stress and $\mathbf{g} = (0, 0, -g)^T$ represents the gravity forces. 121The internal specific (i.e. per mass unit) energy is denoted by e, the temperature by T. 122The heat flux Q_T obeys the Fourier law $Q_T = -\lambda \nabla T$, hence the name "Navier-Stokes-123 Fourier", λ being the heat conductivity. The quantity ∇ denotes $\nabla = \left(\frac{\partial}{\partial x}, \frac{\partial}{\partial y}, \frac{\partial}{\partial z}\right)^T$. 124In the following, we will also use the notations **u** and $\nabla_{x,y}$, $\mathbf{u}(t, x, y, z) = (u, v)^T$ is the 125horizontal velocity and $\nabla_{x,y}$ corresponds to the projection of ∇ on the horizontal plane 126 i.e. $\nabla_{x,y} = \left(\frac{\partial}{\partial x}, \frac{\partial}{\partial y}\right)^T$. The square norm of the velocity vector is $|\mathbf{U}|^2 = u^2 + v^2 + w^2$. 127

We consider a free surface flow (see Fig. 1), therefore we assume

$$z_b(x,y) \le z \le \eta(t,x,y) := h(t,x,y) + z_b(x,y)$$

128 with $z_b(x, y)$ the bottom elevation and h(t, x, y) the water depth.

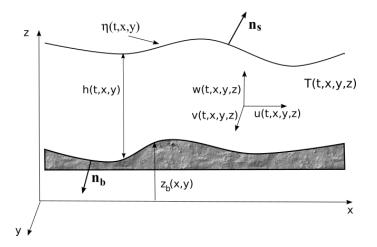


FIG. 1. Flow domain with water height h(t, x, y), free surface $\eta(t, x, y)$ and bottom $z_b(x, y)$.

Notice that, because of the free surface and of the gravitational forces, the term $\rho \mathbf{g}.\mathbf{U} = -\rho g w$ in the right hand side of (2.3) prevents this equation from being directly a local energy conservation law. Nevertheless one can write (see Remark 2.1)

$$\rho g w = \frac{\partial (\rho g z)}{\partial t} + \nabla .(\rho g z \mathbf{U}),$$

which leads to a conservative version of (2.3). In the sequel, we will mainly work with the energy balance written under the form (2.3).

131 Regarding constitutive equations, we assume that the fluid is Newtonian i.e. the 132 viscous part of the Cauchy stress depends linearly on the velocity gradient. Hence 133 the stress tensor Σ is given by

134 (2.4)
$$\boldsymbol{\Sigma} \equiv -p\mathbf{1} + \boldsymbol{\sigma} = -p\mathbf{1} + \boldsymbol{\zeta} \nabla \cdot \mathbf{U}\mathbf{1} + 2\mu D(\mathbf{U}).$$

where μ is the viscosity coefficient, ζ is the second viscosity and $D(\mathbf{U}) = (\nabla \mathbf{U} + (\nabla \mathbf{U})^T)/2$. Further on we shall use the notation A : B for the scalar product between two square matrices A, B, and $|A|^2 = A : A$.

138 Among the thermodynamic variables ρ , p, T, e, only two of them are independent. 139 This implies in particular that we have a state law under the form

140 (2.5)
$$f(\rho, T, p) = 0.$$

141 The thermodynamic variables are linked by the identity

142 (2.6)
$$de = \frac{p}{\rho^2} d\rho + T ds,$$

143 where s is the specific entropy of the fluid. Classically, in order to have a good entropy

structure one has to assume that -s is a convex function of $1/\rho$, e. In Section 3.2 the

145 case for which there is an additional thermodynamic variable S, the specific salinity, 146 is described.

Energy equations can be deduced from the above equations. Multiplying (2.2) by
U yields the kinetic energy equation

149 (2.7)
$$\frac{\partial}{\partial t} \left(\rho \frac{|\mathbf{U}|^2}{2} \right) + \nabla \cdot \left(\left(\rho \frac{|\mathbf{U}|^2}{2} + p - \sigma \right) \mathbf{U} \right) = p \nabla \cdot \mathbf{U} - \sigma : D(\mathbf{U}) + \rho \mathbf{g} \cdot \mathbf{U}$$

150 Subtracting (2.7) to (2.3) gives the equation for the internal energy

151
$$\frac{\partial \rho e}{\partial t} + \nabla \cdot (\rho e \mathbf{U}) = -p \nabla \cdot \mathbf{U} + \sigma : D(\mathbf{U}) - \nabla \cdot Q_T,$$

152 or equivalently

153 (2.8)
$$\rho \frac{De}{Dt} = -p\nabla \cdot \mathbf{U} + \sigma : D(\mathbf{U}) - \nabla \cdot Q_T,$$

with the classical notation $D/Dt \equiv \partial/\partial t + \mathbf{U} \cdot \nabla$. We can write the continuity equation (2.1) as

156 (2.9)
$$\rho \frac{D\rho}{Dt} + \rho^2 \nabla \cdot \mathbf{U} = 0.$$

157 With the thermodynamic relation (2.6) one can write $ds = de/T - (p/T\rho^2)d\rho$, thus 158 multiplying (2.8) by 1/T and (2.9) by $-p/T\rho^2$ we obtain

159
$$\rho \frac{Ds}{Dt} = \frac{1}{T}\sigma : D(\mathbf{U}) - \frac{1}{T}\nabla \cdot Q_T$$

160 This can be written also

161 (2.10)
$$\frac{\partial \rho s}{\partial t} + \nabla \cdot (\rho s \mathbf{U}) = \frac{1}{T} \sigma : D(\mathbf{U}) - \nabla \cdot \frac{Q_T}{T} - Q_T \cdot \frac{\nabla T}{T^2},$$

162 which gives the increase with time of $\int \rho s$, the second principle of thermodynamics.

163 **2.2. Boundary conditions.**

2.2.1. Bottom and free surface. Let \mathbf{n}_b and \mathbf{n}_s be the unit outward normals at the bottom and at the free surface respectively, defined by (see Fig 1)

$$\mathbf{n}_b = \frac{1}{\sqrt{1 + |\nabla_{x,y} z_b|^2}} \left(\begin{array}{c} \nabla_{x,y} z_b \\ -1 \end{array} \right), \qquad \mathbf{n}_s = \frac{1}{\sqrt{1 + |\nabla_{x,y} \eta|^2}} \left(\begin{array}{c} -\nabla_{x,y} \eta \\ 1 \end{array} \right).$$

164 On the bottom we prescribe an impermeability condition

165 (2.11)
$$\mathbf{U} \cdot \mathbf{n}_b = 0,$$

166 and a friction condition given e.g. by a Navier law

167 (2.12)
$$(\boldsymbol{\Sigma} \cdot \mathbf{n}_b) \cdot \mathbf{t}_i = -\kappa \mathbf{U} \cdot \mathbf{t}_i, \quad i = 1, 2$$

168 with κ a Navier coefficient and $(\mathbf{t}_i, i = 1, 2)$ two tangential vectors. For some ap-

169 plications, we rather use more specific friction laws and the equation (2.12) is then 170 replaced by

(
$$\Sigma \cdot \mathbf{n}_b$$
) $\cdot \mathbf{t}_i = -\kappa(h, \mathbf{U}) \cdot \mathbf{t}_i, \quad i = 1, 2,$

172 with $\kappa(h, \mathbf{U}) \cdot \mathbf{U} \geq 0$. On the free surface, we use the kinematic boundary condition

173 (2.13)
$$\frac{\partial \eta}{\partial t} + \mathbf{u}(t, x, y, \eta) \cdot \nabla_{x, y} \eta - w(t, x, y, \eta) = 0,$$

174 and the dynamic boundary condition

175 (2.14)
$$\boldsymbol{\Sigma} \cdot \mathbf{n}_s = -p^a(t, x, y)\mathbf{n}_s + W(t, x, y)\mathbf{t}_W,$$

where $p^a(t, x, y)$, W(t, x, y) are two given quantities, p^a (resp. W) mimics the effects of the atmospheric pressure (resp. the wind blowing at the free surface) and \mathbf{t}_W is a unit horizontal vector given by the wind direction. Throughout the paper $p^a = cst$, W = 0. For the temperature, Neumann or Dirichlet boundary conditions can be taken, see Subsection 2.2.3.

Remark 2.1. Computing the quantity

$$\int_{z_b}^{\eta} z\left(\frac{\partial\rho}{\partial t} + \nabla\cdot\left(\rho\mathbf{U}\right)\right) dz,$$

and using the boundary conditions (2.13), (2.11) one finds

182
$$\frac{\partial}{\partial t} \int_{z_b}^{\eta} g\rho z dz + \nabla_{x,y} \cdot \int_{z_b}^{\eta} g\rho z \mathbf{u} dz = \int_{z_b}^{\eta} g\rho w dz,$$

183 which is the integrated local conservation of gravitational potential energy.

184 **2.2.2. Fluid boundaries and solid walls.** On solid walls we prescribe a non-185 penetration condition (neglecting viscosity)

186
$$\mathbf{U} \cdot \mathbf{n} = 0$$

n being the outward normal to the considered wall, together with appropriate condi tions for the temperature.

In this paper we consider fluid boundaries on which we prescribe zero, one or two of the following conditions depending on the type of the flow (fluvial or torrential): water level $h + z_b(x, y)$ given, discharge $h\mathbf{U} \cdot \mathbf{n}$ given.

192 **2.2.3.** Boundary conditions for the temperature.

The heat flux in Eq. (2.3) requires to define boundary conditions for the temperature. Moreover when the state law (2.5) will be precised, the definition of the temperature at each boundary will be mandatory. We can choose either Neumann or Dirichlet conditions namely at the bottom

197 (2.15)
$$\lambda \frac{\partial T}{\partial \mathbf{n}_b} = \phi_T|_{z_b},$$

199 (2.16)
$$T_b = T_b^0$$
,

200 and at the free surface

201 (2.17)
$$\lambda \frac{\partial T}{\partial \mathbf{n}_s} = \phi_T|_s$$

202 or

203 (2.18)
$$T_s = T_s^0,$$

where $\phi_T|_{z_b}$, $\phi_T|_s$ are two given temperature fluxes and T_b^0 , T_s^0 are two given temperatures.

The system is completed with some initial conditions

$$h(0, x, y) = h^{0}(x, y), \quad \rho(0, x, y) = \rho^{0}(x, y), \quad \mathbf{U}(0, x, y, z) = \mathbf{U}^{0}(x, y, z).$$

3. The low Mach limit. In this section, the low Mach limit of the compressible Navier-Stokes equations is performed in order to get an incompressible system. As already mentioned in the introduction, one of the motivations for this limit is that the density of the water varies very little with pressure variations, and removing acoustic waves from the model is advantageous from the computational point of view.

3.1. The Navier-Stokes-Fourier with temperature. Therefore, we now consider the state equation of the fluid (2.5) under the form

213 (3.1)
$$f(\rho, T, \varepsilon(p - p_{ref})) = 0,$$

where $\varepsilon \ll 1$ is a small parameter (related to the sound velocity, see Remark 3.1) and with p_{ref} a reference pressure constant in space and time. In other words, taking into account the implicit function theorem, this means to assume the particular form for the pressure

218 (3.2)
$$p = p_{ref} + \frac{p_0}{\varepsilon},$$

219 where the law $p_0(\rho, T)$ has no stiff scale.

Remark 3.1. When writing equation (3.1), we assume that the density of the water depends very weakly on the pressure, and this is true in practice. Water and seawater compressibility values at various temperatures and pressures are given in [15, 33]. A possible state law for seawater (involving the salinity *S*, which we will include later on in Section 3.2) is to write (3.1) as

225
$$\rho(S,T,p) = \frac{\rho(S,T,p_{ref})}{1 - \frac{p - p_{ref}}{K(S,T,p - p_{ref})}},$$

where in the fraction $(p - p_{ref})/K(S, T, p - p_{ref})$, the denominator is very large with respect to the numerator, so that this law could actually be written

(3.3)
$$\rho(S,T,p) = \frac{\rho(S,T,p_{ref})}{1 - \varepsilon(p - p_{ref})}.$$

This law was published in [37], where values of the density ρ at different pressures and constant S, T are also given. One can see that the density varies slowly with respect to the pressure. As reported in Appendix A of [26], for S = 8 PSU and T = 10 °C, the water density for p = 0 (atmospheric pressure) is $\rho(8, 10, 0) = 1005.945659$ kg.m⁻³ while for p = 10 bar, the density is $\rho(8, 10, 10) = 1006.41797$ kg.m⁻³. The sound velocity is defined by

$$c^2 = \left(\frac{\partial p}{\partial \rho}\right)_s,$$

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implying after some computations involving relations (2.6) and (3.3) that

$$\varepsilon \approx \frac{1}{\rho(S, T, p_{ref})c^2}.$$

Since in sea water, $c \approx 1500 \text{ m.s}^{-1}$, this justifies the assumption $\varepsilon \ll 1$ done in 229 230 relation (3.1).

Remark 3.2. One could consider that the reference pressure p_{ref} varies in time, for 231instance because of changes in the boundary conditions of the system - p_{ref} adapts to 232 temperature fluxes and mass fluxes at the boundaries. Here, for the sake of simplicity, 233 we consider that p_{ref} is constant in space and in time. Yet the model derivation should 234not be significantly different with $p_{ref} = p_{ref}(t)$. 235

For the thermodynamic identity (2.6) to be compatible with (3.2), it is necessary 236 237to consider a rescaling for e and s, leading to the following rescaled thermodynamic identities 238

239 (3.4)
$$e + \frac{p_{ref}}{\rho} = \frac{e_0}{\varepsilon}, \qquad s = \frac{s_0}{\varepsilon}, \quad \text{with} \quad de_0 = \frac{p_0}{\rho^2} d\rho + T ds_0,$$

where e_0 and s_0 do not involve stiff scales. The low Mach limit is performed by letting 240 ε go to 0 in the Navier-Stokes-Fourier system with the relations (3.2), (3.4). As p is 241the physical pressure, it has to remain finite. Therefore according to (3.2) at the limit 242243 we get $p_0(\rho, T) = 0$. This can be written

244
$$T = T^{eq}(\rho),$$

or equivalently $\rho = \rho(T^{eq})$. In all the remainder of the paper, the superscript e^{eq} (as 245in e_0^{eq}) is used for the quantities "at equilibrium", i.e. quantities in which ρ and T 246 are constrained by the relation $p_0(\rho, T) = 0$. 247

One can define the specific enthalpy $H = e + p/\rho$, and the specific heat capacity 248 at constant pressure $c_p = \left(\frac{\partial H}{\partial T}\right)_p$. Then with (3.2), (3.4) one has $H = H_0/\varepsilon$ with $H_0 = e_0 + p_0/\rho$, and $c_p = c_{p_0}/\varepsilon$ with $c_{p_0} = \left(\frac{\partial H_0}{\partial T}\right)_{p_0}$. We have the following result 2500 -

252**PROPOSITION 3.3.** The system

253 (3.5)
$$\nabla \cdot \mathbf{U} = -\frac{\rho'(T^{eq})}{\rho^2 c_p} \nabla \cdot (\lambda \nabla T^{eq}),$$

254 (3.6)
$$\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mathbf{U}) = 0,$$

255 (3.7)
$$\frac{\partial(\rho \mathbf{U})}{\partial t} + \nabla \cdot (\rho \mathbf{U} \otimes \mathbf{U}) + \nabla p - \nabla \cdot \sigma = \rho \mathbf{g},$$

with the relation $T = T^{eq}(\rho)$, $\rho'(T) = \frac{d\rho}{dT}$ and where p is a Lagrange multiplier, is 256the formal limit of the system (2.1)-(2.3), with (3.2), (3.4) as ε goes to 0. The energy 257balance verified by (3.5)-(3.7) is 258259

260 (3.8)
$$\frac{\partial}{\partial t} \left(\rho \frac{|\mathbf{U}|^2}{2} - p_{ref} + \rho \frac{e_0^{eq}}{\varepsilon} \right) + \nabla \cdot \left(\left(\rho \frac{|\mathbf{U}|^2}{2} - p_{ref} + \rho \frac{e_0^{eq}}{\varepsilon} + p - \sigma \right) \mathbf{U} \right)$$

261
262
$$= \nabla \cdot \left(\frac{\lambda_0}{\varepsilon} \nabla T^{eq} \right) + \rho \mathbf{g} \cdot \mathbf{U} + (p - p_{ref}) \nabla \cdot \mathbf{U} - \sigma : D(\mathbf{U}).$$

The energy balance (3.8) is expressed in the rescaled variables defined by (3.2), (3.4) in order to show clearly the order of magnitude of each term. The quantity λ_0 is defined by (3.11).

266 *Proof.* We first rewrite (2.1) under the form

267 (3.9)
$$\rho \frac{D\rho}{Dt} = -\rho^2 \nabla \cdot \mathbf{U}.$$

268 When $\varepsilon \to 0$ we get $\rho = \rho(T^{eq})$, and we can multiply (3.9) by $dT^{eq}/d\rho$ to get an 269 equation for the temperature

270 (3.10)
$$\rho \frac{DT^{eq}}{Dt} = -\rho^2 \frac{dT^{eq}}{d\rho} \nabla \cdot \mathbf{U}.$$

In the sequel, we consider that for the heat conduction λ and the fluid viscosity μ , we are in the following asymptotic regime,

273 (3.11)
$$\lambda = \frac{\lambda_0}{\varepsilon}, \text{ and } \mu \sim 1.$$

Notice that if λ or μ is smaller than these scales, then the corresponding terms in the incompressible system (3.5)-(3.8) will simply vanish. Hence, the computation of (2.8)- $p_{ref}/\rho^2 \times (2.9)$, multiplying by ε and taking the limit, we get according to (3.4)

277 (3.12)
$$\rho \frac{De_0^{eq}}{Dt} = \nabla \cdot (\lambda_0 \nabla T^{eq}).$$

Writing then as above the rescaled enthalpy $H_0 = e_0 + p_0/\rho$, one gets that at equilibrium

280
$$H_0^{eq} = e_0^{eq},$$

and $H_0^{eq} = H_0^{eq}(T^{eq})$. One has then according to the definition of c_{p_0}

$$dH_0^{eq} = c_{p_0} dT^{eq}$$

283 We thus get a second equation for the temperature

284 (3.13)
$$\rho c_{p_0} \frac{DT^{eq}}{Dt} = \nabla \cdot (\lambda_0 \nabla T^{eq}).$$

Comparing (3.10) and (3.13) we obtain (3.5), where we have replaced the scaled quantities c_{p_0} and λ_0 by their physical values εc_p and $\varepsilon \lambda$ respectively. The pressure p in (3.7) can finally be interpreted as a Lagrange multiplier for the equation (3.5). The momentum equation (3.7) together with the mass equation (3.9) gives again the kinetic energy equation (2.7). Adding it to (3.12) divided by ε and to trivial terms in p_{ref} finally gives the energy balance (3.8).

Remark 3.4. At the limit, the thermodynamic identity (3.4) becomes

$$de_0^{eq} = T^{eq} ds_0^{eq}.$$

From (3.12) we obtain the equation for the evolution of the entropy

292
$$\frac{\partial}{\partial t}(\rho s_0^{eq}) + \nabla \cdot (\rho s_0^{eq} \mathbf{U}) - \frac{1}{T^{eq}} \nabla \cdot (\lambda_0 \nabla T^{eq}) = 0.$$

293 Written in the conservative/dissipative form, this gives

294
$$\frac{\partial}{\partial t}(\rho s_0^{eq}) + \nabla \cdot (\rho s_0^{eq} \mathbf{U}) - \nabla \cdot \left(\lambda_0 \frac{\nabla T^{eq}}{T^{eq}}\right) = \lambda_0 \frac{|\nabla T^{eq}|^2}{(T^{eq})^2},$$

which shows that in accordance with the second law of thermodynamics, the total entropy $\int \rho s_0^{eq}$ can only increase.

The energy balance (3.8) exhibits discrepancies of the order ε with respect to the original energy balance (2.3). Namely, the terms $(p - p_{ref})\nabla \cdot \mathbf{U} - \sigma : D(\mathbf{U})$ are of size ε with respect to the leading order terms and they are not present in (2.3). In order to get an energy balance closer to the original compressible equations, we go one step further and some corrections of size ε are incorporated into the system. These do not modify the order of accuracy of the approximation however, since other error terms of order ε are not corrected.

304 PROPOSITION 3.5. The system

305 (3.14)
$$\nabla \cdot \mathbf{U} = -\frac{\rho'(T^{eq})}{\rho^2 c_p} (\nabla \cdot (\lambda \nabla T^{eq}) + \sigma : D(\mathbf{U})),$$

306 (3.15)
$$\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mathbf{U}) = 0,$$

307 (3.16)
$$\frac{\partial(\rho \mathbf{U})}{\partial t} + \nabla \cdot (\rho \mathbf{U} \otimes \mathbf{U}) + \nabla p - \nabla \cdot \sigma = \rho \mathbf{g},$$

308 with T a given function of ρ

309 (3.17)
$$T = T^{eq}(\rho),$$

ລ.

and where p is a Lagrange multiplier, is an approximation of order ε of the formal limit (3.5)-(3.7) of the system (2.1)-(2.3) with (3.2), (3.4). The system (3.14)-(3.16) satisfies the energy balance equation

313 (3.18)
$$\frac{\partial}{\partial t} \left(\rho \frac{|\mathbf{U}|^2}{2} - p_{ref} + \rho \frac{e_0^c}{\varepsilon} \right) + \nabla \cdot \left(\left(\rho \frac{|\mathbf{U}|^2}{2} - p_{ref} + \rho \frac{e_0^c}{\varepsilon} + p - \sigma \right) \mathbf{U} \right) \\= \nabla \cdot \left(\frac{\lambda_0}{\varepsilon} \nabla T^{eq} \right) + \rho \mathbf{g} \cdot \mathbf{U},$$

314 where e_0^c is an independent (rescaled) corrected energy variable such that $e_0^c - e_0^{eq} =$ 315 $O(\varepsilon)$.

Proof. The proof is similar to that of Proposition 3.3. At the limit, the variables are constrained by the relation (3.17). To obtain an energy balance close to (2.3), we define an independent corrected energy variable e_0^c ($e_0^c \neq e_0^{eq}$) satisfying

319 (3.19)
$$\rho \frac{De_0^c}{Dt} = \nabla \cdot (\lambda_0 \nabla T^{eq}) - \varepsilon (p - p_{ref}) \nabla \cdot \mathbf{U} + \varepsilon \sigma : D(\mathbf{U}).$$

With this correction, the energy balance is now (3.18). The internal energy at equilibrium e_0^{eq} satisfies (3.19) but without the term $\varepsilon(p - p_{ref})\nabla \cdot \mathbf{U}$, which implies that $e_0^c - e_0^{eq}$ is of order ε . Recalling that $H_0^{eq} = e_0^{eq}$, we get the equation for the enthalpy at equilibrium H_0^{eq}

324 (3.20)
$$\rho \frac{DH_0^{eq}}{Dt} = \nabla \cdot (\lambda_0 \nabla T^{eq}) + \varepsilon \sigma : D(\mathbf{U}).$$

Eq. (3.20) appears as a correction of (3.12). Notice that if the correction $-\varepsilon(p - p_{ref})\nabla \cdot \mathbf{U}$ were incorporated, the model obtained would be even more accurate, but it would contain derivatives of p, which makes the equations much more difficult to handle since p is a Lagrange multiplier. Here we restrict ourselves to the correction $\varepsilon \sigma : D(\mathbf{U})$, which is viscous dissipation, to keep the model simple.

From equation (3.20), we get a corrected equation on the temperature

331
$$\rho c_{p_0} \frac{DT^{eq}}{Dt} = \nabla \cdot (\lambda_0 \nabla T^{eq}) + \varepsilon \sigma : D(\mathbf{U}).$$

Consequently, comparing with (3.10), we get the equation on $\nabla \cdot \mathbf{U}$ (in the rescaled variables)

334
$$\nabla \cdot \mathbf{U} = -\frac{\rho'(T^{eq})}{\rho^2 c_{p_0}} (\nabla \cdot (\lambda_0 \nabla T^{eq}) + \varepsilon \sigma : D(\mathbf{U})),$$

which is (3.14). In physical variables, the equation for the temperature can also be written

337 (3.21)
$$\rho c_p \frac{DT^{eq}}{Dt} = -\nabla \cdot Q_T + \sigma : D(\mathbf{U}).$$

Remark 3.6. The entropy equation for the model (3.14)-(3.16) is

339
$$\frac{\partial}{\partial t}(\rho s_0^{eq}) + \nabla \cdot (\rho s_0^{eq} \mathbf{U}) - \nabla \cdot \left(\lambda_0 \frac{\nabla T^{eq}}{T^{eq}}\right) = \lambda_0 \frac{|\nabla T^{eq}|^2}{(T^{eq})^2} + \frac{\varepsilon}{T^{eq}}\sigma : D(\mathbf{U}).$$

As $\sigma : D(\mathbf{U}) \ge 0$, we obtain again that the total entropy $\int \rho s_0^{eq}$ can only increase. Here there is no discrepancy with the original entropy equation (2.10).

Remark 3.7. In models (3.5)-(3.7) and (3.14)-(3.16), the temperature T^{eq} is no longer an independent variable of the system. It is recovered by inverting the state law $\rho = \rho(T^{eq})$.

Remark 3.8. The model (3.14)-(3.16) is very similar to what is classically obtained 345 in the literature when the low-Mach limit of the Navier-Stokes equations with thermal 346 conduction is taken, see for instance [30]. The limit is usually performed by expanding 347 the variables of the system in power series of the Mach number. The method we have 348 used here is different. As we have given ourselves only a generic state law, we cannot 349 express the Mach number, let alone make it appear in the equations. Instead, the 350 result is obtained via a rescaling of the variable part of the pressure, and this enables 351 to consider general state laws. 352

Remark 3.9. Even if a gas and water behave very differently, an example of equation of state to which the previous asymptotics can be applied is the stiffened gas law [19]

$$p = (\gamma - 1)\rho e - \gamma p_{\infty}, \quad \frac{c_p}{\gamma}T = e - \frac{p_{\infty}}{\rho},$$

with the constraint $e - p_{\infty}/\rho > 0$, where $\gamma > 1$, $p_{\infty} > 0$, $c_p > 0$ are constants. Here the entropy is given by $s = \frac{c_p}{\gamma} \log(T/\rho^{\gamma-1})$. Then the scaling assumptions (3.2), (3.4) are satisfied when

360
$$p_{\infty} = -p_{ref} + \frac{p_{\infty 0}}{\varepsilon}, \qquad c_p = \frac{c_{p_0}}{\varepsilon},$$

356

with $p_{\infty 0}$ and c_{p_0} constants independent of ε . At equilibrium we get the relation $\rho T^{eq} = \frac{\gamma}{\gamma - 1} \frac{p_{\infty 0}}{c_{p_0}}$, thus T^{eq} is inversely proportional to ρ .

0,

Remark 3.10. In [6] the authors focused on the following 2D (x, z) model

364 (3.22)
$$\frac{\partial \rho}{\partial t} + \frac{\partial \rho u}{\partial x} + \frac{\partial \rho w}{\partial z} =$$

365 (3.23)
$$\frac{\partial \rho u}{\partial t} + \frac{\partial \rho u^2}{\partial x} + \frac{\partial \rho u w}{\partial z} + \frac{\partial p}{\partial x} = \frac{\partial \sigma_{xx}}{\partial x} + \frac{\partial \sigma_{xz}}{\partial z}$$

366 (3.24)
$$\frac{\partial p}{\partial z} = -\rho g + \frac{\partial \sigma_{zx}}{\partial x} + \frac{\partial \sigma_{zz}}{\partial z}$$

367 (3.25) $\rho = \rho(T),$

368 (3.26)
$$\frac{\partial \rho T}{\partial t} + \frac{\partial \rho u T}{\partial x} + \frac{\partial \rho w T}{\partial z} = \frac{\lambda}{c_p} \frac{\partial^2 T}{\partial x^2} + \frac{\lambda}{c_p} \frac{\partial^2 T}{\partial z^2}.$$

Rewriting equation (3.26) in the non-conservative form, we get

370
$$\rho \frac{\partial T}{\partial t} + \rho u \frac{\partial T}{\partial x} + \rho w \frac{\partial T}{\partial z} = \frac{\lambda}{c_p} \left(\frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial z^2} \right).$$

371 Multiplying this equation by $\rho'(T)/\rho$ gives an equation for ρ

372 (3.27)
$$\frac{\partial \rho}{\partial t} + u \frac{\partial \rho}{\partial x} + w \frac{\partial \rho}{\partial z} = \frac{\rho'(T)}{\rho} \frac{\lambda}{c_p} \left(\frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial z^2} \right)$$

Finally, subtracting (3.22) to (3.27) and rearranging the terms gives a compatibility condition similar to (3.5) for a constant λ

375
$$\frac{\partial u}{\partial x} + \frac{\partial w}{\partial z} = -\frac{\rho'(T)}{\rho^2 c_p} \lambda \left(\frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial z^2} \right).$$

Indeed (3.26) is similar to (3.13). This shows that the model (3.22)-(3.26) corresponds to the model (3.5)-(3.7) with a hydrostatic assumption, see paragraph 4.

Because of the stability inherited from the energy balance (3.18), that is consistent with (2.3), in the sequel we consider the system (3.14)-(3.16) instead of (3.5)-(3.7).

380 **3.2. The Navier-Stokes-Fourier system with salinity.** We now consider the 381 situation where the fluid density depends on the temperature T and on another inter-382 nal variable, the specific salinity S. This is the case of sea water. The compressible 383 Navier-Stokes-Fourier system with temperature and salinity can be written

384 (3.28)
$$\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mathbf{U}) = 0,$$

385 (3.29)
$$\frac{\partial(\rho \mathbf{U})}{\partial t} + \nabla \cdot (\rho \mathbf{U} \otimes \mathbf{U}) + \nabla p - \nabla \cdot \sigma = \rho \mathbf{g},$$

386 (3.30)
$$\frac{\partial}{\partial t} \left(\rho \frac{|\mathbf{U}|^2}{2} + \rho e \right) + \nabla \cdot \left(\left(\rho \frac{|\mathbf{U}|^2}{2} + \rho e + p - \sigma \right) \mathbf{U} \right) = -\nabla \cdot \mathbf{F}^T + \rho \mathbf{g} \cdot \mathbf{U},$$

387 (3.31) $\frac{\partial(\rho S)}{\partial t} + \nabla \cdot (\rho S \mathbf{U}) = -\nabla \cdot \mathbf{F}^S.$

The local mass and momentum conservation equations are identical to (2.1) and (2.2), whereas the energy equation is slightly modified: the heat flux is now \mathbf{F}^T . The 390 conservation equation (3.31) on the mass fraction of chlorides S can also be written 391 as

392 (3.32)
$$\rho \frac{DS}{Dt} = -\nabla \cdot \mathbf{F}^S,$$

with \mathbf{F}^{S} the salt flux. According to [38], the molecular fluxes of heat and salt \mathbf{F}^{T} and \mathbf{F}^{S} are expressed in terms of the thermodynamic Onsager forces related to the entropy equation (3.38) below,

396 (3.33)
$$\mathbf{F}^{S} = A\nabla\left(\frac{-\mu_{S}}{T}\right) + B\nabla\left(\frac{1}{T}\right),$$

397 (3.34)
$$\mathbf{F}^{T} = B\nabla\left(\frac{-\mu_{S}}{T}\right) + C\nabla\left(\frac{1}{T}\right)$$

where A, B and C are three independent coefficients to be discussed later, and μ_S is the chemical potential of seawater. The particular symmetric form of (3.33), (3.34) is related to Onsager's reciprocity principle and the dissipation of entropy, see (3.39)

401 below. The state equation of the fluid is

402
$$f(\rho, T, S, p) = 0,$$

403 and the thermodynamic identity now reads

404 (3.35)
$$de = \frac{p}{\rho^2} d\rho + T ds + \mu_S dS.$$

As in classical gas dynamics, a simple assumption that ensures the hyperbolic structure of the model is that -s is a convex function of $1/\rho, e, S$. From (3.30) and using (3.29), (3.28) we get the equation on the internal energy

408 (3.36)
$$\rho \frac{De}{Dt} = -p\nabla \cdot \mathbf{U} + \sigma : D(\mathbf{U}) - \nabla \cdot \mathbf{F}^T.$$

Let us explain how the formulas (3.33), (3.34) lead to the second law of thermodynamics. The equation for the entropy is obtained using the thermodynamic identity (3.35) combined with the mass, internal energy and salinity equations (2.9), (3.36), (3.32),

413 (3.37)
$$\rho\left(\frac{\partial s}{\partial t} + \mathbf{U} \cdot \nabla s\right) = \frac{1}{T} \left(\sigma : D(\mathbf{U}) - \nabla \cdot \mathbf{F}^T + \mu_S \nabla \cdot \mathbf{F}^S\right),$$

 $^{414}_{415}$ that can be written under conservative/dissipative form $^{415}_{415}$

416 (3.38)
$$\frac{\partial \rho s}{\partial t} + \nabla \cdot (\rho s \mathbf{U}) = \frac{1}{T} \sigma : D(\mathbf{U}) - \nabla \cdot \left(\frac{1}{T} \mathbf{F}^T - \frac{\mu_S}{T} \mathbf{F}^S\right) + \mathbf{F}^T \cdot \nabla \left(\frac{1}{T}\right) - \mathbf{F}^S \cdot \nabla \left(\frac{\mu_S}{T}\right).$$

Substituting the expressions (3.33), (3.34) in the right-hand side of (3.38), we obtain the following quadratic form for the nonconservative terms in the right-hand side of (3.38), (3.38),

423 (3.39)
$$\mathbf{F}^T \cdot \nabla\left(\frac{1}{T}\right) - \mathbf{F}^S \cdot \nabla\left(\frac{\mu_S}{T}\right) = C \left|\nabla\left(\frac{1}{T}\right)\right|^2$$

424
425 $-2B\nabla\left(\frac{1}{T}\right) \cdot \nabla\left(\frac{\mu_S}{T}\right) + A \left|\nabla\left(\frac{\mu_S}{T}\right)\right|^2.$

the three constraints are A > 0, C > 0 and $AC > B^2$. With these constraints the 427 expressions (3.33), (3.34) of \mathbf{F}^{S} and \mathbf{F}^{T} can be written in terms of the gradients of 428

the salinity S, temperature T and pressure p (as in [38], equations (B.26) and (B.27)) 429

by writing $\mu_S = \mu_S(T, S, p)$ and assuming $\partial_S \mu_S > 0$, 430

431 (3.40)
$$\mathbf{F}^{S} = -\rho k^{S} \left(\nabla S + \frac{\partial_{p} \mu_{S}}{\partial_{S} \mu_{S}} \nabla p \right) - \left(\frac{\rho k^{S} T}{\partial_{S} \mu_{S}} \partial_{T} \left(\frac{\mu_{S}}{T} \right) + \frac{B}{T^{2}} \right) \nabla T,$$

432 (3.41)
$$\mathbf{F}^T = -\rho c_p k^T \nabla T + \frac{B \partial_S \mu_S}{\rho k^S T} \mathbf{F}^S,$$

where $k^T > 0$ and $k^S > 0$ are the thermal and molecular diffusivities of salt, related 433 to A, B, C by 434

435 (3.42)
$$A = \frac{\rho k^S T}{\partial_S \mu_S}, \qquad C = \rho c_p k^T T^2 + \frac{B^2}{A}.$$

The free model coefficients are thus now k^S , k^T and B. Note that \mathbf{F}^T in (3.41) is 436 written as a gradient of T (as in the case where the temperature is the only tracer), 437 plus another term, due to the presence of salt. Using (3.42) and (3.33), the quadratic 438form (3.39) can be rewritten 439

440

14

441 (3.43)
$$C \left| \nabla \left(\frac{1}{T} \right) \right|^2 - 2B\nabla \left(\frac{1}{T} \right) \cdot \nabla \left(\frac{\mu_S}{T} \right) + A \left| \nabla \left(\frac{\mu_S}{T} \right) \right|^2$$

442
443 $= \rho c_p k^T T^2 \left| \nabla \frac{1}{T} \right|^2 + \frac{1}{A} \left| \mathbf{F}^S \right|^2,$

which shows that it is indeed nonnegative. Thus with (3.38) the total entropy $\int \rho s$ 444 can only increase, in accordance with the second principle of thermodynamics. 445 446

We now perform the low Mach limit as in Section 3. We introduce the state 447 equation of the fluid under the form 448

449 (3.44)
$$\tilde{f}(\rho, T, S, \varepsilon(p - p_{ref})) = 0,$$

with $\varepsilon \ll 1$. Taking into account the thermodynamic identity (3.35), p, e, s are thus 450rescaled as in (3.2), (3.4) and μ_S scales as $1/\varepsilon$, which yields 451

452 (3.45)
$$p = p_{ref} + \frac{p_0}{\varepsilon}, \quad e + \frac{p_{ref}}{\rho} = \frac{e_0}{\varepsilon}, \quad s = \frac{s_0}{\varepsilon}, \quad \mu_S = \frac{\mu_{S0}}{\varepsilon},$$

with $de_0 = \frac{p_0}{\rho^2} d\rho + T ds_0 + \mu_{S0} dS$. We still define the specific enthalpy $H = e + p/\rho$, and 453the specific heat capacity at constant pressure $c_p = \left(\frac{\partial H}{\partial T}\right)_{p,S}$. Then one has $H = H_0/\varepsilon$ 454with $H_0 = e_0 + p_0/\rho$, and $c_p = c_{p_0}/\varepsilon$ with $c_{p_0} = \left(\frac{\partial H_0}{\partial T}\right)_{p_0,S}$. As $\varepsilon \to 0$, the finiteness of p yields the equilibrium relation $p_0(\rho, T, S) = 0$ or equivalently $T = T^{eq}(\rho, S)$. 455456

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457 PROPOSITION 3.11. The system

$$\begin{aligned} \nabla\cdot\mathbf{U} &= \frac{1}{\rho^2 c_p} \left(\frac{\partial\rho}{\partial T^{eq}}\right)_S \left((T^{eq})^2 \left(\frac{\partial\left(\mu_S/T^{eq}\right)}{\partial T^{eq}}\right)_S \nabla\cdot\mathbf{F}^S + \nabla\cdot\mathbf{F}^T \\ &-\sigma: D(\mathbf{U}) + \frac{1}{\rho^2} \left(\frac{\partial\rho}{\partial S}\right)_{T^{eq}} \nabla\cdot\mathbf{F}^S, \end{aligned}$$

460 (3.47) $\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mathbf{U}) = 0,$

461 (3.48)
$$\frac{\partial(\rho \mathbf{U})}{\partial t} + \nabla \cdot (\rho \mathbf{U} \otimes \mathbf{U}) + \nabla p - \nabla \cdot \sigma = \rho \mathbf{g},$$

462 (3.49)
$$\frac{\partial(\rho S)}{\partial t} + \nabla \cdot (\rho S \mathbf{U}) = -\nabla \cdot \mathbf{F}^{S},$$

463 (3.50)
$$\mathbf{F}^{S} = -\rho k^{S} \nabla S - \left(\frac{\rho k^{S} T}{\partial_{S} \mu_{S}} \partial_{T} \left(\frac{\mu_{S}}{T}\right) + \frac{B}{T^{2}}\right) \nabla T,$$

464 (3.51)
$$\mathbf{F}^T = -\rho c_p k^T \nabla T + \frac{B \partial_S \mu_S}{\rho k^S T} \mathbf{F}^S,$$

with $T = T^{eq}(\rho, S)$ and where p is a Lagrange multiplier, is an approximation of order 66 ε of the formal limit of the system (3.28)-(3.31), (3.40), (3.41), (3.45).

467 Proof. At the limit there remain only two independent thermodynamic variables, 468 and one can take T and S. We consider that \mathbf{F}^S is bounded but $\mathbf{F}^T \sim 1/\varepsilon$. With 469 (3.33), (3.34) this means that $A \sim \varepsilon$, $B \sim 1$, $C \sim 1/\varepsilon$, and using (3.42) that

470 (3.52)
$$k^S \sim 1, \quad c_p k^T \sim 1/\varepsilon.$$

471 At equilibrium the term in ∇p in (3.40) disappears in the expression of \mathbf{F}^{S} since μ_{S} 472 depends only weakly on p (this is a consequence of the scaling assumption (3.44)) and 473 ∇p remains bounded, giving (3.50), (3.51). The mass conservation equation (3.28), the 474 momentum equation (3.29) and the salinity equation (3.31) are unchanged. Consider 475 then the enthalpy $H = e + p/\rho = H_0/\varepsilon$. Taking (3.36) $-p_{ref}/\rho^2 \times (2.9)$, multiplying 476 by ε and taking the limit, we get according to (3.45) and using the equilibrium $H_0^{eq} = e_0^{eq}$,

478 (3.53)
$$\rho \frac{DH_0^{eq}}{Dt} = -\varepsilon \nabla \cdot \mathbf{F}^T + \varepsilon \sigma : D(\mathbf{U}).$$

479 One can write at equilibrium

480 (3.54)
$$dH^{eq} = \left(\frac{\partial H^{eq}}{\partial T^{eq}}\right)_S dT^{eq} + \left(\frac{\partial H^{eq}}{\partial S}\right)_{T^{eq}} dS.$$

481 Combining (3.53) (written in the physical variables) with (3.54) gives

482 (3.55)
$$\rho\left(\frac{\partial H^{eq}}{\partial T^{eq}}\right)_{S}\frac{DT^{eq}}{Dt} - \left(\frac{\partial H^{eq}}{\partial S}\right)_{T^{eq}}\nabla\cdot\mathbf{F}^{S} + \nabla\cdot\mathbf{F}^{T} - \sigma: D(\mathbf{U}) = 0.$$

483 We have similarly for the density

484 (3.56)
$$d\rho = \left(\frac{\partial\rho}{\partial T^{eq}}\right)_S dT^{eq} + \left(\frac{\partial\rho}{\partial S}\right)_{T^{eq}} dS.$$

485 The quantities $\left(\frac{\partial \rho}{\partial T^{eq}}\right)_S$ and $\left(\frac{\partial \rho}{\partial S}\right)_{T^{eq}}$ are known from the state law of salted water, 486 and using (3.9), (3.32) we deduce another equation on the temperature

487 (3.57)
$$-\rho^2 \nabla \cdot \mathbf{U} = \left(\frac{\partial \rho}{\partial T^{eq}}\right)_S \rho \frac{DT^{eq}}{Dt} - \left(\frac{\partial \rho}{\partial S}\right)_{T^{eq}} \nabla \cdot \mathbf{F}^S.$$

488 Combining (3.55) with (3.57) gives an expression for $\rho^2 \nabla \cdot \mathbf{U}$ 489

490 (3.58)
$$\rho^{2}\nabla\cdot\mathbf{U} = \frac{\left(\frac{\partial\rho}{\partial T^{eq}}\right)_{S}}{\left(\frac{\partial H^{eq}}{\partial T^{eq}}\right)_{S}} \left(-\left(\frac{\partial H^{eq}}{\partial S}\right)_{T^{eq}}\nabla\cdot\mathbf{F}^{S} + \nabla\cdot\mathbf{F}^{T} - \sigma:D(\mathbf{U})\right) + \left(\frac{\partial\rho}{\partial S}\right)_{T^{eq}}\nabla\cdot\mathbf{F}^{S}$$

that generalizes (3.14). We recall that by definition $c_p = \left(\frac{\partial H}{\partial T}\right)_{p,S}$, thus $c_p = c_{p_0}/\varepsilon$ and at equilibrium $c_{p_0} = \left(\frac{\partial H_0}{\partial T^{eq}}\right)_S$, or in physical variables

495 (3.59)
$$c_p = \left(\frac{\partial H^{eq}}{\partial T^{eq}}\right)_S,$$

which enables to express the denominator in (3.58). Note that we obtain (3.58) without using the thermodynamic identity (3.35), and without involving μ_S . Next in (3.58) it remains to express $\partial H^{eq}/\partial S$. One has at equilibrium (see [38], equations (A.11.1) and (A.11.2))

500 (3.60)
$$\left(\frac{\partial H^{eq}}{\partial S}\right)_{T^{eq}} = \mu_S - T^{eq} \frac{\partial \mu_S}{\partial T^{eq}} = -(T^{eq})^2 \frac{\partial}{\partial T^{eq}} (\mu_S/T^{eq}),$$

thus finally (3.58) gives (3.46). The relation (3.60) can be deduced from the limit of the thermodynamic identity (3.45). At equilibrium we have

503 (3.61)
$$ds_0^{eq} = \frac{dH_0^{eq}}{T^{eq}} - \frac{\mu_{S0}}{T^{eq}}dS,$$

that can be interpreted as $dH^{eq} = T^{eq}ds^{eq} + \mu_S dS$, analogous to (3.35). We reformulate (3.61) as

506
$$d\left(s_{0}^{eq} - \frac{H_{0}^{eq}}{T^{eq}}\right) = \frac{H_{0}^{eq}}{(T^{eq})^{2}}dT^{eq} - \frac{\mu_{S0}}{T^{eq}}dS$$

507 The left-hand side is an exact differential form, therefore we can write that the two 508 cross derivatives with respect to T, S and S, T are equal. It yields

509
$$\frac{\partial}{\partial T^{eq}} \left(-\frac{\mu_{S0}}{T^{eq}} \right) = \frac{\partial}{\partial S} \left(\frac{H_0^{eq}}{(T^{eq})^2} \right),$$

510 which gives (3.60).

511 Remark 3.12. Because of (3.61), (3.53), the entropy equation (3.38) is still valid 512 for our model (3.46)-(3.51), and the quadratic form on the right-hand side takes the 513 form (3.43).

16

514 Remark 3.13. A criterion of well-posedness of our incompressible system (3.46)-515 (3.51) can be derived as follows. We write that the second-order terms in the coupled 516 S and T^{eq} equations (3.49), (3.55) give a diffusion matrix with positive eigenvalues. 517 With (3.50), (3.51) we have at equilibrium (using (3.60))

518 (3.62)
$$\mathbf{F}^{S} = -\rho k^{S} \nabla S - E \nabla T, \quad \text{with } E = \frac{\rho k^{S} T}{\partial_{S} \mu_{S}} \partial_{T} \left(\frac{\mu_{S}}{T}\right) + \frac{B}{T^{2}},$$

519

520 (3.63)
$$\mathbf{F}^{T} - \left(\frac{\partial H^{eq}}{\partial S}\right)_{T} \mathbf{F}^{S} = -\rho c_{p} k^{T} \nabla T + \left(\frac{B \partial_{S} \mu_{S}}{\rho k^{S} T} + T^{2} \partial_{T} (\mu_{S}/T)\right) \mathbf{F}^{S}$$
$$= -\rho c_{p} k^{T} \nabla T + E \frac{T \partial_{S} \mu_{S}}{\rho k^{S}} \mathbf{F}^{S}.$$

521 The diffusion matrix of the system is thus, taking into account (3.59),

522
$$\begin{pmatrix} \rho k^S & E \\ E \frac{T \partial_S \mu_S}{c_p} & E^2 \frac{T \partial_S \mu_S}{\rho k^S c_p} + \rho k^T \end{pmatrix}.$$

523 We obtain positive eigenvalues under the natural conditions also mentioned in [38]

524 (3.64)
$$k^S > 0, \quad k^T > 0, \quad \partial_S \mu_S > 0, \quad c_p > 0$$

525 Remark 3.14. For the particular choice of B such that E = 0 (the relation between 526 E and B is (3.62)) we have a diagonal diffusion matrix in S, T. In this case we have 527 particular formulas

528 (3.65)
$$\mathbf{F}^S = -\rho k^S \nabla S,$$

529 and from (3.63), (3.60)

530 (3.66)
$$\mathbf{F}^T + T^2 \partial_T \left(\frac{\mu_S}{T}\right) \mathbf{F}^S = -\rho c_p k^T \nabla T.$$

531 We deduce that

532 (3.67)
$$\nabla \cdot \mathbf{F}^T + T^2 \partial_T \left(\frac{\mu_S}{T}\right) \nabla \cdot \mathbf{F}^S = -\nabla \cdot \left(\rho c_p k^T \nabla T\right) - \mathbf{F}^S \cdot \nabla \left(T^2 \partial_T \left(\frac{\mu_S}{T}\right)\right),$$

533 that can be used in (3.46). Next we write (3.55) using (3.59), (3.60) and (3.67) to get

534 (3.68)
$$\rho c_p \frac{DT}{Dt} = \nabla \cdot \left(\rho c_p k^T \nabla T\right) + \mathbf{F}^S \cdot \nabla \left(T^2 \partial_T \left(\frac{\mu_S}{T}\right)\right) + \sigma : D(\mathbf{U})$$

535 We can write

$$\nabla \left(T^2 \partial_T \left(\frac{\mu_S}{T} \right) \right) = \nabla \left(-\mu_S + T \partial_T \mu_S \right)$$

= $-\partial_T \mu_S \nabla T - \partial_S \mu_S \nabla S + \nabla \left(T \partial_T \mu_S \right)$
= $-\partial_S \mu_S \nabla S + T \nabla \left(\partial_T \mu_S \right),$

536

538 (3.69)
$$\rho c_p \frac{DT}{Dt} = \nabla \cdot \left(\rho c_p k^T \nabla T\right) + T \mathbf{F}^S \cdot \nabla \left(\partial_T \mu_S\right) - \partial_S \mu_S \mathbf{F}^S \cdot \nabla S + \sigma : D(\mathbf{U}),$$

539 that generalizes (3.21). We therefore obtain the equations (3.49) and (3.69) on S and

- 540 T, with the value of \mathbf{F}^{S} given by (3.65). With the conditions (3.64) we can see that
- 541 T remains nonnegative and S verifies the maximum principle (i.e. it remains within
- its initial lower and upper bounds). Notice that with (3.59), (3.60), the model finally relies on the knowledge of $\rho^{eq}(T,S)$, $H^{eq}(T,S)$, $k^T(T,S)$, $k^S(T,S)$.

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4. Hydrostatic approximation of the Navier-Stokes-Fourier system. We consider the incompressible Navier-Stokes-Fourier system (3.14)-(3.16) derived in paragraph 3, for the sake of simplicity we consider that the density only depends on the temperature $\rho = \rho(T)$, and we neglect the salinity S. For the sake of lightness, the exponent ^{eq} is dropped in this part and in the rest of the present document. The system is completed with the boundary conditions (2.11)-(2.14).

550 The hydrostatic assumption consists in neglecting the vertical acceleration of the 551 fluid

552 (4.1)
$$\rho\left(\frac{\partial w}{\partial t} + \frac{\partial uw}{\partial x} + \frac{\partial vw}{\partial y} + \frac{\partial w^2}{\partial z}\right) \approx 0,$$

where we recall that $\mathbf{U} = (\mathbf{u}, w)$ with $\mathbf{u} = (u, v)$. We refer to [10, 17, 25] for the analysis of hydrostatic models and for their asymptotic derivation [5, 14, 24].

555 4.1. The incompressible and hydrostatic Navier-Stokes-Fourier system.

Using the assumption (4.1) and the definition (2.4), the hydrostatic approximation of the incompressible Navier-Stokes-Fourier system (3.14)-(3.16) therefore reads

558
$$\nabla \cdot \mathbf{U} = -\frac{\rho'(T)}{\rho^2 c_p} \left(\nabla \cdot (\lambda \nabla T) + \zeta \left(\nabla_{x,y} \cdot \mathbf{u} + \partial_z w \right)^2 \right)$$

559 (4.2)
$$+2\mu\Big(|D_{x,y}\mathbf{u}|^2 + \frac{1}{2}|\nabla_{x,y}w + \partial_z\mathbf{u}|^2 + (\partial_zw)^2\Big)\Big),$$

560 (4.3)
$$\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mathbf{U}) = 0,$$

561 $\frac{\partial (\rho \mathbf{u})}{\partial t} + \nabla_{x,y} \cdot (\rho \mathbf{u} \otimes \mathbf{u}) + \frac{\partial (\rho \mathbf{u} w)}{\partial z} + \nabla_{x,y} p = \nabla_{x,y} \Big(\zeta \big(\nabla_{x,y} \cdot \mathbf{u} + \partial_z w \big) \Big)$

562 (4.4)
$$+\nabla_{x,y} \cdot \left(2\mu D_{x,y}\mathbf{u}\right) + \partial_z \left(\mu \left(\nabla_{x,y}w + \partial_z \mathbf{u}\right)\right),$$

with $T = T(\rho)$ a given function. According to Remark 3.2, p_{ref} is taken to be the atmospheric pressure p^a that is supposed constant.

567 The energy balance for the hydrostatic system (4.2)-(4.5) is

$$\frac{\partial}{\partial t} \left(\rho \frac{|\mathbf{u}|^2}{2} + \rho e^c \right) + \nabla \cdot \left(\left(\rho \frac{|\mathbf{u}|^2}{2} + \rho e^c + p - p^a \right) \mathbf{U} \right)$$

568
$$+ \nabla_{x,y} \cdot \left(-\zeta \left(\nabla_{x,y} \cdot \mathbf{u} + \partial_z w \right) \mathbf{u} - 2\mu (D_{x,y} \mathbf{u}) \mathbf{u} - \mu \left(\nabla_{x,y} w + \partial_z \mathbf{u} \right) w \right)$$

$$+ \partial_z \left(-\zeta \left(\nabla_{x,y} \cdot \mathbf{u} + \partial_z w \right) w - \mu \left(\nabla_{x,y} w + \partial_z \mathbf{u} \right) \cdot \mathbf{u} - 2\mu w \partial_z w \right) = \nabla \cdot (\lambda \nabla T) - \rho g w,$$

569 with the corrected internal energy e^c governed by

570 (4.6)
$$\frac{\partial}{\partial t}(\rho e^{c}) + \nabla \cdot (\rho e^{c}\mathbf{U}) = -(p - p^{a})\left(\nabla_{x,y} \cdot \mathbf{u} + \partial_{z}w\right) + \nabla \cdot (\lambda \nabla T) + \zeta\left(\nabla_{x,y} \cdot \mathbf{u} + \partial_{z}w\right)^{2} + 2\mu\left(|D_{x,y}\mathbf{u}|^{2} + \frac{1}{2}|\nabla_{x,y}w + \partial_{z}\mathbf{u}|^{2} + (\partial_{z}w)^{2}\right).$$

19

Notice that (4.6) is similar to (3.19) divided by ε . The energy e(T) defined by $de/dT = c_p$ (e corresponds to e^{eq} in paragraph 3) satisfies the same equation (4.6) except the term $-(p - p^a)\nabla \cdot \mathbf{U}$ in the right-hand side. It is obtained by multiplying (4.3) by $c_p dT/d\rho$ and by using (4.2).

In order to make the numerical approximation easier and without loss of applicability we propose a few usual simplifications for the viscous terms. First, let us notice that for water the second viscosity cannot be neglected compared to the dynamic viscosity, see [12]. But in practice, the chosen value for the coefficient μ (typically $\mu \approx 0.1 \text{ kg.m}^{-1}.\text{s}^{-1}$) is much bigger than the value of the dynamic viscosity ($\approx 10^{-3}$ kg.m⁻¹.s⁻¹) in order to model the turbulence effects. Hence hereafter, we can neglect the second viscosity i.e. we consider $\zeta = 0$.

Notice that we have from Eq. (4.2) $\nabla \cdot \mathbf{U} = \mathcal{O}((\lambda + \mu)\rho'/(\rho^2 c_p))$ and therefore the terms in $\mathcal{O}(\mu(\lambda + \mu)\rho'/(\rho^2 c_p)) \ll 1$ appearing in Eqs. (4.2),(4.4) can be neglected. Likewise, neglecting the viscous terms in Eq. (4.5) and in the third component of (2.14), an integration between z and η of (4.5) gives the relation

586 (4.7)
$$p = p^a + \int_z^{\eta} \rho g dz_1.$$

Hence we propose to replace the system (4.2)-(4.5) by the following simplified system with constant μ

589 (4.8)
$$\nabla \cdot \mathbf{U} = -\frac{\rho'(T)}{\rho^2 c_p} \left(\nabla \cdot (\lambda \nabla T) + \mu \left| \nabla_{x,y} \mathbf{u} \right|^2 + \mu \left| \frac{\partial \mathbf{u}}{\partial z} \right|^2 \right),$$

590 (4.9)
$$\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mathbf{U}) = 0,$$

591
$$\frac{\partial(\rho \mathbf{u})}{\partial t} + \nabla_{x,y} \cdot (\rho \mathbf{u} \otimes \mathbf{u}) + \frac{\partial(\rho \mathbf{u}w)}{\partial z} + \nabla_{x,y} \int_{z}^{\eta} \rho g dz_{1} = \mu \nabla_{x,y} \cdot \nabla_{x,y} \mathbf{u}$$

592 (4.10)
$$+\mu \frac{\partial^2 \mathbf{u}}{\partial z^2}.$$

593 The system (4.8)-(4.10) is completed with the energy balance

594
$$\frac{\partial}{\partial t} \left(\rho \frac{|\mathbf{u}|^2}{2} + \rho e \right) + \nabla_{x,y} \cdot \left(\mathbf{u} \left(\rho \frac{|\mathbf{u}|^2}{2} + \int_z^\eta \rho g dz_1 + \rho e \right) - \mu \nabla \frac{|\mathbf{u}|^2}{2} \right) = \nabla . (\lambda \nabla T),$$

595 with e is the internal energy of the fluid governed by

596
$$\frac{\partial(\rho e)}{\partial t} + \nabla_{x,y} \cdot (\rho e \mathbf{u}) = \left(\int_{z}^{\eta} \rho g dz_{1}\right) \nabla_{x,y} \cdot \mathbf{u} + \mu |\nabla_{x,y} \mathbf{u}|^{2} + \mu \left|\frac{\partial \mathbf{u}}{\partial z}\right|^{2} + \nabla (\lambda \nabla T).$$

Therefore the formulation of the rheology terms appearing in (4.10) is similar to what has been studied by some of the authors in [3].

For the system (4.8)-(4.10), the boundary conditions (2.11),(2.13) and (2.15)-(2.18) remain valid. Concerning the dynamical boundary conditions for Eq. (4.10), the assumptions used above for the viscous terms imply that the boundary condition (2.12) becomes

603 (4.11)
$$\mu \sqrt{1 + |\nabla_{x,y} z_b|^2} \frac{\partial \mathbf{u}}{\partial \mathbf{n}_b} = -\kappa \mathbf{u},$$

with κ a Navier coefficient. Notice that up to terms in $\mathcal{O}(\mu | \nabla_{x,y} z_b |)$, relations (2.12) and (4.11) are similar. For some applications, one can choose $\kappa = \kappa(h, \mathbf{u}|_b)$.

Likewise, the modified expression of the viscous terms leads to a simplified version of the dynamic boundary condition at the free surface and Eq. (2.14) becomes

608 (4.12)
$$\mu \frac{\partial \tilde{\mathbf{u}}}{\partial \mathbf{n}_s} = W(t, x, y) \mathbf{t}_W.$$

where $\tilde{\mathbf{u}} = (\mathbf{u}, 0)^T$. The third component of (2.14) corresponds to $p_s = p^a$ that has been used to obtain (4.7)

4.2. The hydrostatic Euler-Fourier system. Assuming that the viscous terms
 are small, the incompressible Navier-Stokes-Fourier system (3.14)-(3.16) becomes the
 incompressible Euler-Fourier system

614 (4.13)
$$\nabla \cdot \mathbf{U} = -\frac{\rho'(T)}{\rho^2 c_p} \nabla \cdot (\lambda \nabla T),$$

615 (4.14)
$$\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mathbf{U}) = 0,$$

616 (4.15)
$$\frac{\partial(\rho \mathbf{U})}{\partial t} + \nabla \cdot (\rho \mathbf{U} \otimes \mathbf{U}) + \nabla p = \rho \mathbf{g}.$$

- 617 Therefore, the hydrostatic approximation of the system (4.13)-(4.15) consists in the
- 618 hydrostatic Euler-Fourier model

619 (4.16)
$$\nabla \cdot \mathbf{U} = -\frac{\rho'(T)}{\rho^2 c_p} \nabla \cdot (\lambda \nabla T),$$

620 (4.17)
$$\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mathbf{U}) = 0,$$

621 (4.18)
$$\frac{\partial(\rho \mathbf{u})}{\partial t} + \nabla_{x,y} \cdot (\rho \mathbf{u} \otimes \mathbf{u}) + \frac{\partial(\rho \mathbf{u}w)}{\partial z} + \nabla_{x,y} \int_{z}^{\eta} \rho g dz = 0,$$

completed with the relation $\rho = \rho(T)$ and the boundary conditions (2.11), (2.13).

4.3. Remark on the Boussinesq assumption. In geophysical water flows, density variations are often considered as small and this allows justifying the Boussinesq assumption, which consists in considering the density variations only in the gravitational forces. More precisely, assuming

$$\rho = \rho(T) = \rho_0 + f(T),$$

with $f(T) \ll \rho_0$, this leads to writing the incompressible hydrostatic Euler system (4.16)-(4.18) under the form

625 (4.19) $\nabla \cdot \mathbf{U} = 0,$

626 (4.20)
$$\rho_0 c_p \left(\frac{\partial T}{\partial t} + \mathbf{U} \cdot \nabla T\right) = \nabla \cdot (\lambda \nabla T),$$

627 (4.21)
$$\rho_0\left(\frac{\partial \mathbf{u}}{\partial t} + \nabla_{x,y} \cdot (\mathbf{u} \otimes \mathbf{u}) + \frac{\partial(\mathbf{u}w)}{\partial z}\right) + \nabla_{x,y} \int_z^\eta \rho g dz = 0$$

Notice that whereas in (4.16), the divergence of the velocity field equals the expansion due to the temperature effects, the Boussinesq assumption implies the divergence free condition (4.19).

The Boussinesq assumption is valid in various regimes [27, 16] but

21

- 632 it does not ensure a conservation of the kinetic energy since $\rho_0 \frac{|\mathbf{u}|^2}{2}$ is conserved 633 instead of $\rho \frac{|\mathbf{u}|^2}{2}$,
- for long time phenomena (sloshing, wave propagation,...) significant differences appear when the Boussinesq assumption is made, see [6, paragraph 6.2].

In this work, the Boussinesq assumption is not done and some remarks about its validity are given in the forthcoming paper [7], see also [27, 16].

5. Conclusion. In this paper we have derived, using a low-Mach approxima-638 tion, an incompressible model for variable density flows. This model is obtained from 639 640 compressible Navier-Stokes equations by assuming that the density depends weakly on the pressure, thus involving a small parameter ε , that can eventually be identified 641 642 as the Mach number. As $\varepsilon \to 0$, the pressure dependency is lost, while the dependency of the density on temperature and eventually salinity remains. It leads to an 643 "incompressible" model where the divergence of the velocity is equal to source terms 644 645involving temperature and salinity fluxes. In order to obtain a more accurate model in terms of energy balance, a correction of order ε is incorporated, which is indeed the 646 viscous dissipation. The resulting model does not rely on the Boussinesq approxima-647 tion. It is mass-conservative; expansion or contraction can be observed as a result of 648 the variation of a tracer concentration. Our models generalize classical incompressible 649 models to quite general state laws that can depend on salinity. They include entropy 650 consistency via Onsager's principles. 651

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