# About dynamical system modelling Illustration in micro-biology

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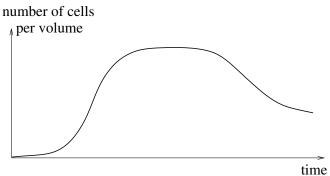
#### **Contents**

- Modeling bacterial growth in batch and continuous cultures
- Simple representations of space
- Questions related to biodiversity
- Modeling and analysis of growth inhibition
- Facing models with experimental data

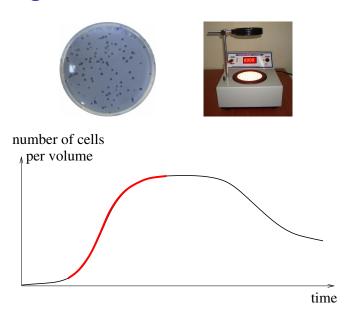
# **Bacterial growth in batch**



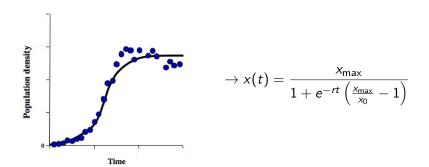




# **Bacterial growth in batch**



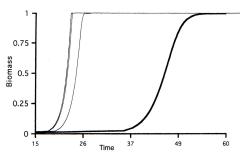
# The logistic growth (Verhulst, 1838)



Equivalent formulation:  $x(\cdot)$  is solution of the differential equation

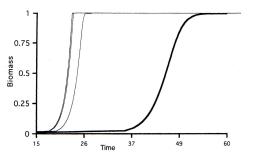
$$\frac{dx}{dt} = rx\left(1 - \frac{x}{x_{\text{max}}}\right)$$

# Jacques Monod's experiments (1930)





# Jacques Monod's experiments (1930)





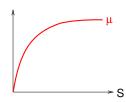
does not fit the logistic curves...

# Mathematical modeling

**Hypothesis H0.** There exists y s.t. x + ys = m = cste

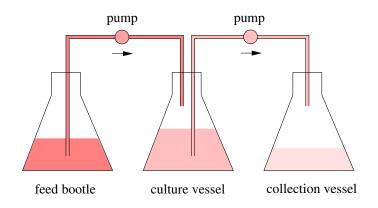
**Hypothesis H1a.** 
$$\frac{dx}{dt} = \mu sx \Rightarrow \frac{dx}{dt} = \underbrace{\mu \frac{m}{y}}_{r} x \left(1 - \frac{x}{m}\right)$$

**Hypothesis H1b.**  $\frac{dx}{dt} = \mu(s)x$  where  $\mu(\cdot)$  is not necessarily linear:



How to identify with accuracy this specific growth curve?

#### The chemostat device



Monod 1950 - Novick & Szilard 1950

# **Dynamical modelling**

▶ Mechanics: 2nd motion law:

$$mass imes acceleration = \sum forces$$

Bio-reaction kinetics: mass balance (for biotic and abiotic):

$$\textit{mass variation} = \sum \textit{inputs} - \sum \textit{outputs} + \underbrace{\textit{growth} - \textit{death}}_{\textit{for biotic only}}$$

# **Dynamical modelling**

chemical kinetics:

Ex.: 
$$\alpha A + \beta B \longrightarrow P = A^{\alpha} B^{\beta} \implies v = \frac{dP}{dt} = k[A]^{\alpha} [B]^{\beta}$$

microbial kinetics:

Ex.: 
$$B + kS \longrightarrow B + B \longrightarrow v = \frac{dB}{dt} = 1815$$

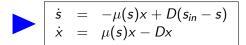
growth is not simply a matter of matching bacteria with molecules of substrate...

# The mathematical model (in concentrations)

$$\begin{bmatrix} \frac{dx}{dt} \\ \frac{ds}{dt} \end{bmatrix} = \begin{bmatrix} \mu(s)x \\ -\frac{\mu(s)}{y}x \end{bmatrix} + \begin{bmatrix} -\frac{Q}{V}x \\ \frac{Q}{V}(s_{in} - s) \end{bmatrix}$$
growth dilution

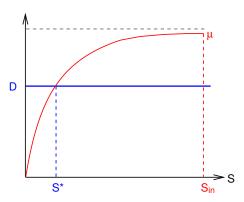
Remark. 
$$Q = 0 \Rightarrow x + ys = constant$$

Simplification and notations. 
$$y = 1$$
  $\dot{} = \frac{d}{dt}$   $D = \frac{Q}{V}$ 



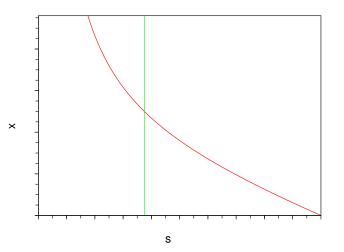
## **Determination of equilibria**

$$\dot{s} = -\mu(s)x + D(s_{in} - s)$$
  $\Rightarrow$   $x^* = 0$  or  $\mu(s^*) = D$   $x^* = s_{in}$   $x^* = s_{in} - s^*$  wash-out positive equilibrium



#### **Null-clines**

$$\dot{s} = -\mu(s)x + D(s_{in} - s)$$
  
$$\dot{x} = \mu(s)x - Dx$$

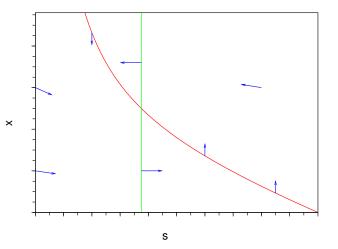


$$\dot{x}=0$$

$$\dot{s} = 0$$

#### **Vector field**

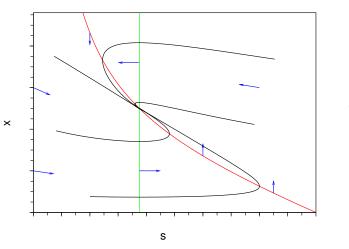
$$\dot{s} = -\mu(s)x + D(s_{in} - s)$$
$$\dot{x} = \mu(s)x - Dx$$



$$\dot{x}=0$$

## Phase portrait

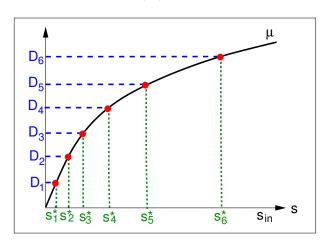
$$\dot{s} = -\mu(s)x + D(s_{in} - s)$$
  
$$\dot{x} = \mu(s)x - Dx$$



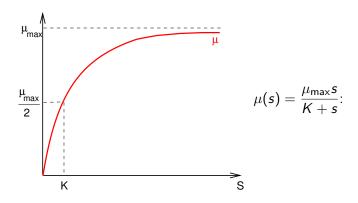
$$\dot{x} = 0$$

### **Back to experiments**

For each dilution rate D, one obtains a steady state  $s^*$  with  $\mu(s^*) = D$ :



#### The Monod's law



#### The chemostat model

$$\dot{s} = -\mu(s)x + D(s_{in} - s)$$
  
$$\dot{x} = \mu(s)x - Dx$$

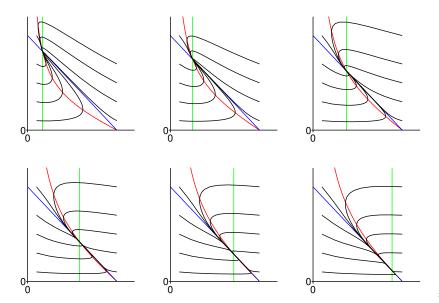


Ecology of mountain lakes

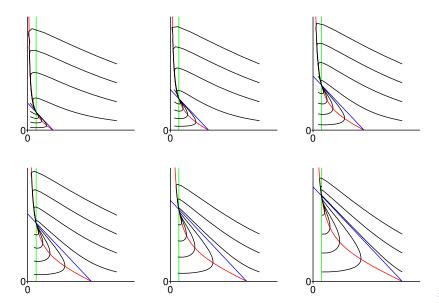


Industrial bioreactors

# For various increasing dilution rates



# For various increasing input concentrations

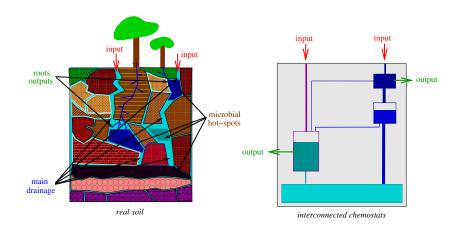


# About conversion yield at equilibrium

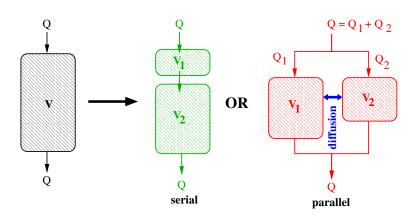
- ▶ The mathematical model of the chemostat predicts that the substrate concentration at equilibrium is independent of the input concentration  $s_{in}$  (provided that  $\mu(s_{in}) > D$ ).
- Micro-biologists report that this property is not verified when the tank is not homogeneous or in natural ecosystems such as soil ecosystems.

**Question:** What is the influence of a spatial repartition on output substrate concentration at steady state?

## The paradigm of simple representations of space

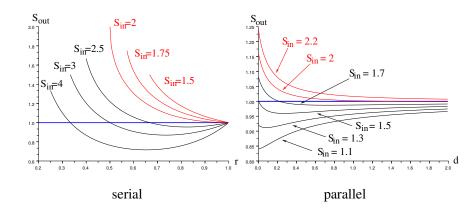


# Study of some simple spatial configurations



with 
$$V = V_1 + V_2$$

## Comparison of performances at steady-state



# Roles of spatial structure and diffusion

**Message 1.** For any monotonic  $\mu(\cdot)$ , there exists a threshold  $\bar{s}_{in}$  such that

- for  $s_{in} > \overline{s}_{in}$ , the serial configuration is more efficient,
- ▶ for  $s_{in} < \overline{s}_{in}$ , the parallel configuration is more efficient,

Message 2. For the parallel configurations,

- for  $s_{in} > \overline{s}_{in}$ , the map  $d \mapsto s_{out}^{\star}(d)$  is decreasing,
- for  $s_{in} < \bar{s}_{in}$ , the map  $d \mapsto s^{\star}_{out}(d)$  admits an unique minimum for  $d^{\star} < +\infty$

Furthermore, there exists another threshold  $\underline{s}_{in} < \overline{s}_{in}$  s.t.  $d^* = 0$  for  $s_{in} < \underline{s}_{in}$ .

## Questions related to microbial biodiversity

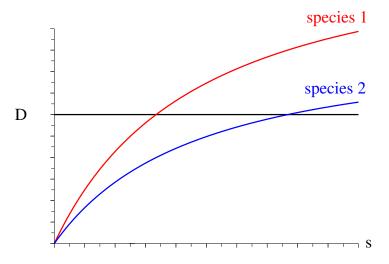
- ► How to model microbial competition?
- Can a biodiversity be maintened?
- Does spatial structures impact biodiversity?
- Is biodiversity is in favor of bioconversion yielding?

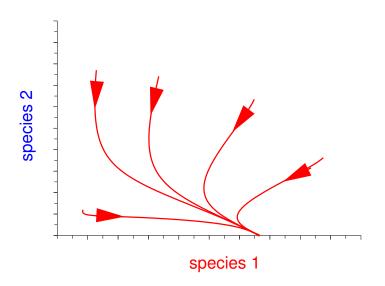
## Having two species in the chemostat

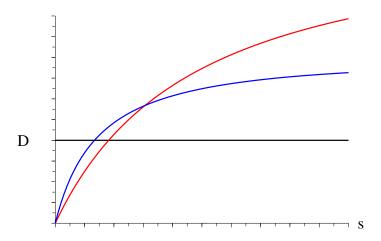
$$\dot{s} = -\mu_1(s)x_1 - \mu_2(s)x_2 + D(s_{in} - s) 
\dot{x}_1 = \mu_1(s)x_1 - Dx_1 
\dot{x}_2 = \mu_2(s)x_2 - Dx_2$$

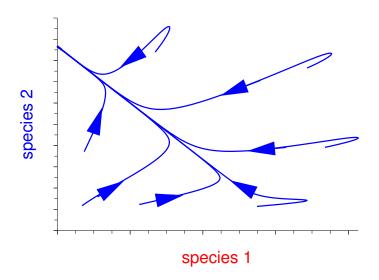
#### Equilibria:

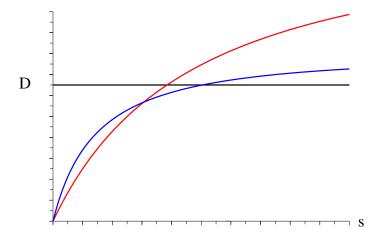
wash-out	species 1 only	species 2 only	species coexistence
$\left[\begin{array}{c} s_{in} \\ 0 \\ 0 \end{array}\right]$	$\left[\begin{array}{c} s_1^\star \\ s_{in} - s_1^\star \\ 0 \end{array}\right]$	$\begin{bmatrix} s_2^{\star} \\ 0 \\ s_{in} - s_2^{\star} \end{bmatrix}$	would require $\mu_1(s^\star) = \mu_2(s^\star) = D$ non generic condition!

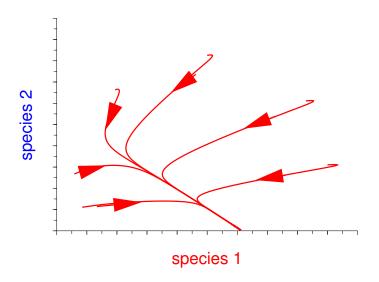








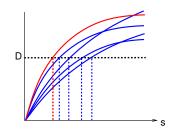




## The Competitive Exclusion Principle

Assume that all  $\mu_i(\cdot)$  are increasing

Let 
$$\Lambda_i(D) = \{s \geq 0 \mid \mu_i(s) > D\}$$
 and define the break-even concentrations  $\lambda_i(D) = \inf \Lambda_i(D)$ 



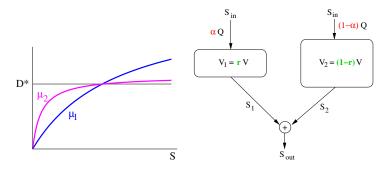
#### Principle.

- Generically, there is at most one species at equilibrium.
- ► The equilibrium (if it exists) with the species that possesses the smallest break-even concentration is the only attractive equilibrium.

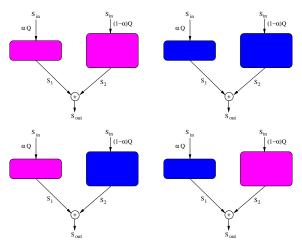
# About niches and over-yielding

Consider two strains:

and the spatial structure:



## Bioconversion and over-yielding



What is the most efficient configuration?

# Overyielding at equilibrium

For increasing  $\mu(\cdot)$ , define the **conditional break-even conc.** 

$$\bar{\lambda}(D) = \begin{vmatrix} \mu^{-1}(D) & \text{if } D \leq \mu(s_{in}) \\ s_{in} & \text{if } D > \mu(s_{in}) \end{vmatrix}$$

and consider 
$$F(\alpha,r) := \alpha \bar{\lambda} \left( \frac{\alpha}{r} D \right) + (1-\alpha) \bar{\lambda} \left( \frac{1-\alpha}{1-r} D \right)$$

Definition: There is over-yielding when

$$G(\alpha, r) := \alpha \bar{\lambda}_1 \left(\frac{\alpha}{r} D\right) + (1 - \alpha) \bar{\lambda}_2 \left(\frac{1 - \alpha}{1 - r} D\right) < \min(F_1(\alpha, r), F_2(\alpha, r))$$

# Overyielding at equilibrium

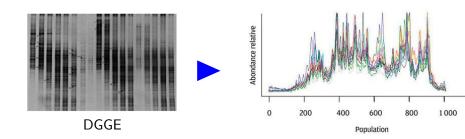
Assume there exists  $D^{\star}$  such that  $\bar{\lambda}_1(D^{\star}) = \bar{\lambda}_2(D^{\star})$ .

Proposition. If  $\mu(\cdot)$  is **concave**, then there exist configurations  $(\alpha, r)$  such that

$$\frac{\alpha}{r}D < D^{\star} < \frac{1-\alpha}{1-r}D < \min(\mu_1(s_{in}), \mu_2(s_{in}))$$

that exhibit over-yielding.

#### The high diversity case



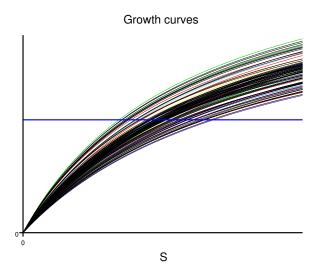
With tools of molecular biology (fingerprints), biologists observe bioreactors with substrate and biomass concentrations at **quasi steady state**, while the **diversity** of the biomass is **high** and its composition is **constantly evolving**...

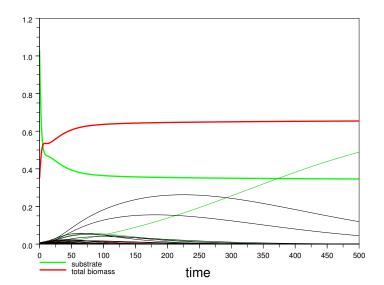
$$\dot{s} = -\sum_{j=1}^{n} \mu_j(s) x_j + D(s_{in} - s)$$

$$\dot{x}_i = \mu_i(s) x_i - Dx_i \qquad (i = 1 \cdots n)$$

#### Numerical simulations with n large (over hundred):

- random drawing of functions  $\mu_i(\cdot)$  among a family of Monod functions
- random drawing of the initial composition of the biomass

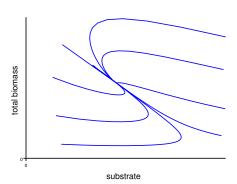




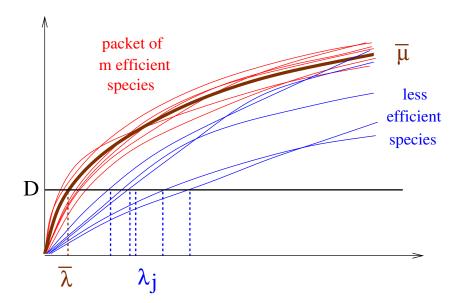
Let 
$$b = \sum_{i=1}^{n} x_i$$
 et  $p_i = \frac{x_i}{b}$ 

$$\Rightarrow \begin{cases} \dot{s} = -\tilde{\mu}(s, \mathbf{p})b + D(s_{in} - s) \\ \dot{b} = \tilde{\mu}(s, \mathbf{p})b - Db \end{cases} \text{ with } \tilde{\mu}(s, \mathbf{p}) = \sum_{i=1}^{n} p_{i}\mu_{i}(s)$$

and 
$$\dot{p}_i = (\mu_i(s) - \tilde{\mu}(s, p)) p_i$$



# Splitting species in two groups



#### The fast dynamics

$$\begin{cases} \dot{s} = -\bar{\mu}(s)\bar{x} - \sum_{j=m+1}^{n} \mu_{j}(s)x_{j} + D(s_{in} - s) \\ \dot{\bar{x}} = \bar{\mu}(s)\bar{x} - D\bar{x} \\ \dot{x}_{j} = \mu_{j}(s)x_{j} - Dx_{j} \qquad (j = m+1 \cdots n) \end{cases}$$

Competitive Exclusion Principle: 
$$\begin{cases} \lim_{t \to +\infty} s(t) = s^* \\ \lim_{t \to +\infty} \bar{x}(t) = x^* = s_{in} - s^* \\ \lim_{t \to +\infty} x_j(t) = 0 \qquad (j = m + 1 \cdots n) \end{cases}$$

## The slow dynamics

for 
$$i=1\cdots m$$
 let write  $\mu_i(s)=\bar{\mu}(s)+\displaystyle\frac{\varepsilon}{\nu_i(s)}$ 

$$\Rightarrow \dot{p}_i = \varepsilon \left( \nu_i(s^*) - \sum_{j=1}^m p_j \nu_j(s^*) \right) p_i$$

Explicit solution:

$$p_i(t) = rac{p_i(0)e^{A_i \mathcal{E} t}}{\displaystyle\sum_{i=1}^m p_j(0)e^{A_j \mathcal{E} t}} \quad ext{with} \quad A_j = 
u_j(s^\star)$$

#### Properties of the slow dynamics

- ▶ Each proportion  $t \mapsto p_i(t)$  is
  - either increasing up to a time  $T_i$  and then decreasing,
  - either increasing for any t,
  - either decreasing for any t.
- ▶ When  $\varepsilon \to 0$  then  $T_i \to +\infty$  for all species excepted one.
- ▶ The "reactional entropy"

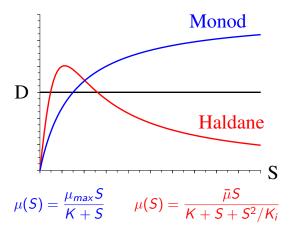
$$E(t) := \sum_{j} \mu_{j}(\bar{s}) p_{j}(t)$$

is increasing toward D.

#### **Concluding remark**

The Competitive Exclusion Principle provides only an asymptotic result.

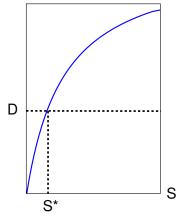
#### Substrate inhibition

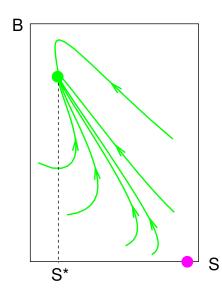


e.g. J. F. Andrews, A mathematical model for the continuous culture of microorganisms utilizing inhibitory substrates, Biotech. Bioengrg., 10 (1968), pp. 707-723.

#### The chemostat model with the Monod law

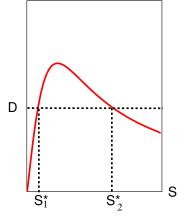
$$\mu(S) = \frac{\mu_{\mathsf{max}} S}{K + S}$$

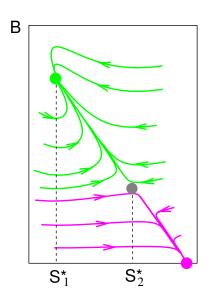




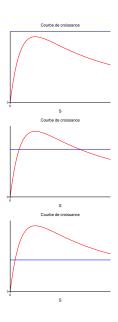
#### The chemostat model with the Haldane law

$$\mu(S) = \frac{\bar{\mu}S}{K + S + S^2/K_i}$$





#### Possible behaviors



$$D > \max_{s \in [0, S_{in}]} \mu(s)$$

1 equilibrium: wash-out

$$\mu(s_{in}) < D < \max_{s \in [0,S_{in}]} \mu(s)$$

3 equilibria : bi-stability

$$D<\mu(s_{in})$$

2 equilibria: stability

## **Operating real bioreactors**



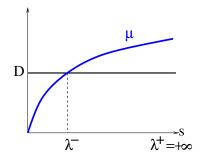
INRA LBE (Narbonne)

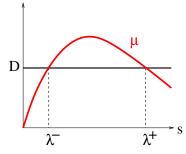
"It happens that *some* additional species need to be added for the bioprocess to **start**, but these additional species are **no longer present at steady state**."

## **Generalization of the Exclusion Principle**

**Hypothesis:**  $E_i(D) = \{s > 0 \mid \mu_i(s) > D\}$  is an interval

$$(\lambda_i^-(D),\lambda_i^+(D))$$





#### **Generalization of the Exclusion Principle**

#### Proposition.

Let 
$$Q(D) = \bigcup_{i} E_{i}(D) = \bigcup_{i} (\lambda_{i}^{-}(D), \lambda_{i}^{+}(D))$$

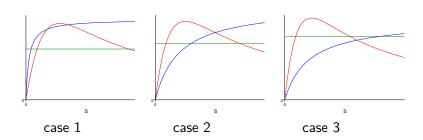
- ▶ There is generically competitive exclusion.
- ▶ There are as many species that could win the competition as the number of connected components of the set Q(D).

## **Adding another species**

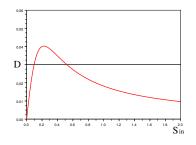
$$\dot{s} = -\mu_1(s)x_1 - \mu_2(s)x_2 + D(s_{in} - s)$$

$$\dot{x}_1 = \mu_1(s)x_1 - Dx_1 \qquad of Haldane type$$

$$\dot{x}_2 = \mu_2(s)x_2 - Dx_2 \qquad of Monod type$$



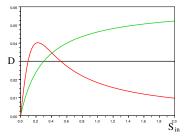
# About the bio-augmentation

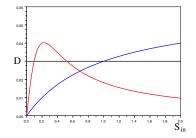


$$\max_{[0,s_{in}]} \mu < \mu(s_{in}) < D$$

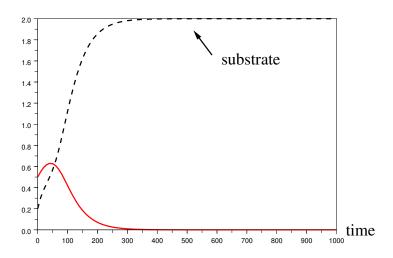
 $\Rightarrow$  the wash-out is attractive...

What is the effect of adding strain "blue" or "green" (in small quantity)?

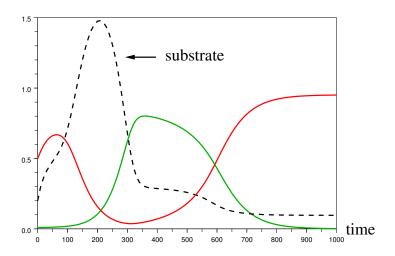




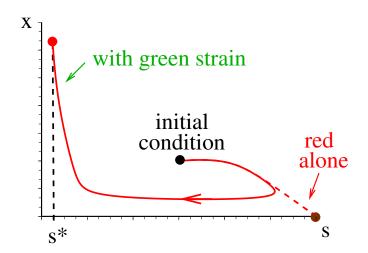
#### Red species alone



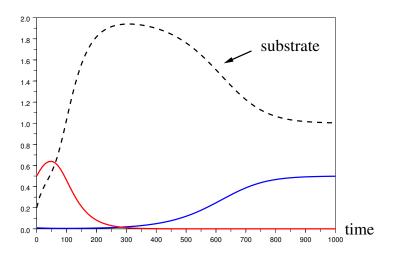
#### Adding green strain



#### Phase portrait



## **Adding blue strain**



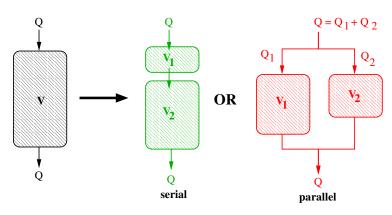
## Toward the "biological control"

Let  $E = (\lambda^-, \lambda^+)$  for the red species. Let  $\lambda$  be the break-even conc. for the additional species.

- $\lambda \in E$ : stabilization of the red species
- ▶  $\lambda \notin E$ : possibility of invasion

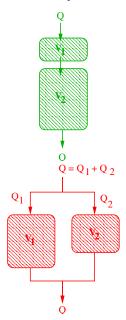
#### Playing with interconnections

Consider 
$$D = \frac{Q}{V}$$
 s.t.  $\mu(s_{in}) < D < \max_{s \in [0, S_{in}]} \mu(s)$ .



with 
$$\boldsymbol{V} = \boldsymbol{V}_1 + \boldsymbol{V}_2$$

#### Serial and parallel patterns



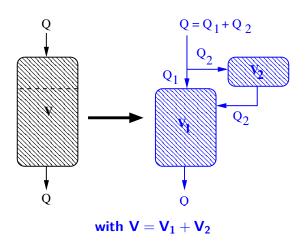
$$V_1 < V \Rightarrow rac{Q}{V_1} > D \,, \quad V_2 < V \Rightarrow rac{Q}{V_2} > D$$

The wash-out equilibrium is attractive in both tanks.

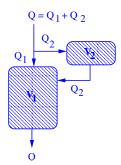
$$rac{Q_1}{V_1} < \mu(\mathit{s_{in}}) \Rightarrow rac{Q_2}{V_2} > \mu(\mathit{s_{in}})$$
 (and vice-versa)

The wash-out equilibrium is attractive in at least one tank.

#### The buffered pattern



#### The buffered chemostat



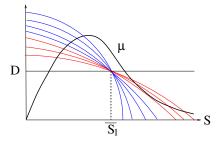
Two parameters:  $(\alpha, r)$  with

$$\frac{Q_2}{V_2} = \alpha D \text{ et } V_1 = {}^{r}V$$

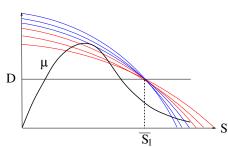
- buffer tank: classical chemostat  $\Rightarrow$  unique positive equ. if  $lpha D < \mu(s_{in})$
- main tank: chemostat with double inputs:

$$\mu(s_1^{\star}) = \frac{D}{r} - \alpha D \left(1 - \frac{1}{r}\right) \frac{s_{in} - s_2^{\star}}{s_{in} - s_1^{\star}}$$

## A graphical characterization



or

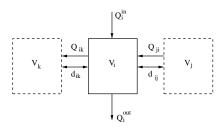


# A matricial generalization

$$\dot{S} = -\frac{1}{y}R(S,X) + MS + DS^{in}$$
$$\dot{X} = R(S,X) + MX + DX^{in}$$

where D is a diagonal matrix, and M is a compartmental matrix:

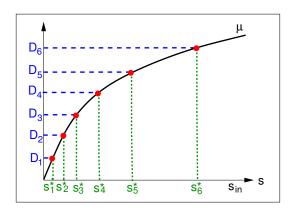
- i.  $M_{ii} \leq 0$  for any indice i,
- ii.  $M_{ij} \ge 0$  for any indices  $i \ne j$ ,
- iii. for any indice i, one has  $\sum M_{ij} \leq 0$



that fulfills the Kirchoff law of mass conservation.

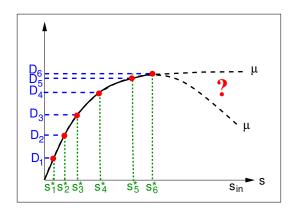
#### **About experimental identification**

*Principle:* one fixes a dilution rate  $D_i$  and "wait" for the corresponding equilibrium  $s_i^*$ .



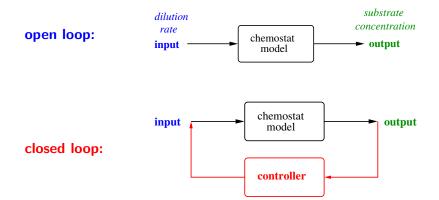
#### **About experimental identification**

*Principle:* one fixes a dilution rate  $D_i$  and "wait" for the corresponding equilibrium  $s_i^*$ .



#### Change of view point

One fixes a reference value  $s^*$  and look for **stabilizing** the chemostat at  $s = s^*$  adjusting the dilution rate D.



# **Adaptive controller**

Fix  $\bar{D}$  as a value of the dilution rate.

▶ Step 1: With the feedback law

$$D(\bar{D},s) = \bar{D} + G_1(s^* - s)$$

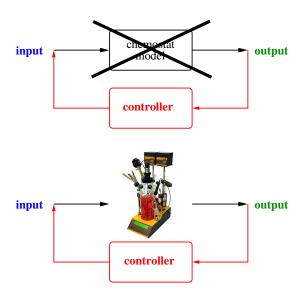
the closed-loop chemostat admits an equilibrium with  $s = \bar{s}$  such that  $\mu(\bar{s}) = D(\bar{D}, \bar{s})$ .

▶ **Step 2:** With the adaptation law for  $\bar{D}$ :

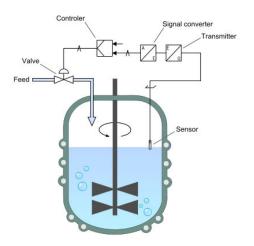
$$\frac{d\bar{D}}{dt}=G_2(s^{\star}-s)$$

the closed-loop chemostat admits the equilibrium with  $s=s^*$  and  $\bar{D}^*=\mu(s^*)$ .

#### Facing the real life...



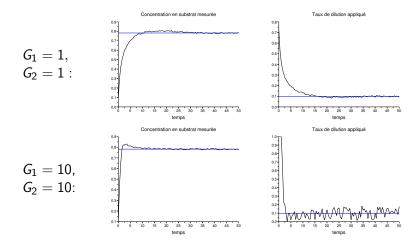
#### The auxostat



see Gostomski, Muhlemann, Lin, Mormino & Bungay. Auxostats for continuous culture research. Journal of Biotechnology, 1994



## Simulations with noisy measurements

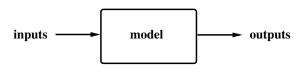




#### **General comments**

- ➤ A model is ONE representation of the reality. There exist modelS.
- A model is generally built for a question of interest
  - ⇒ different representations for the same object.
- A model is a representation ALONG WITH HYPOTHESES
  - $\Rightarrow$  a validity domain is associated to a model.
- ► Simple models can represent complex systems, depending on the scale of representation.

#### Inputs-outputs models



inputs	model	outputs	
known	known	unknown	prediction
known	unknown	known	identification
unknown	known	known	control

## **Further reading**

- ▶ J. HAEFNER. *Modeling Biological Systems*, Springer 2005.
- ► C. TAUBES. Mathematical Differential Equations in Biology, Cambridge 2008.
- ► L. EDELSTEIN-KESHET. *Mathematical Models in Biology*, SIAM 2005.
- ▶ N. Britton. *Essential Mathematical Modelling*, Springer 2005.